

1 **Valorization of coffee cherry waste ash as a sustainable construction** 2 **material**

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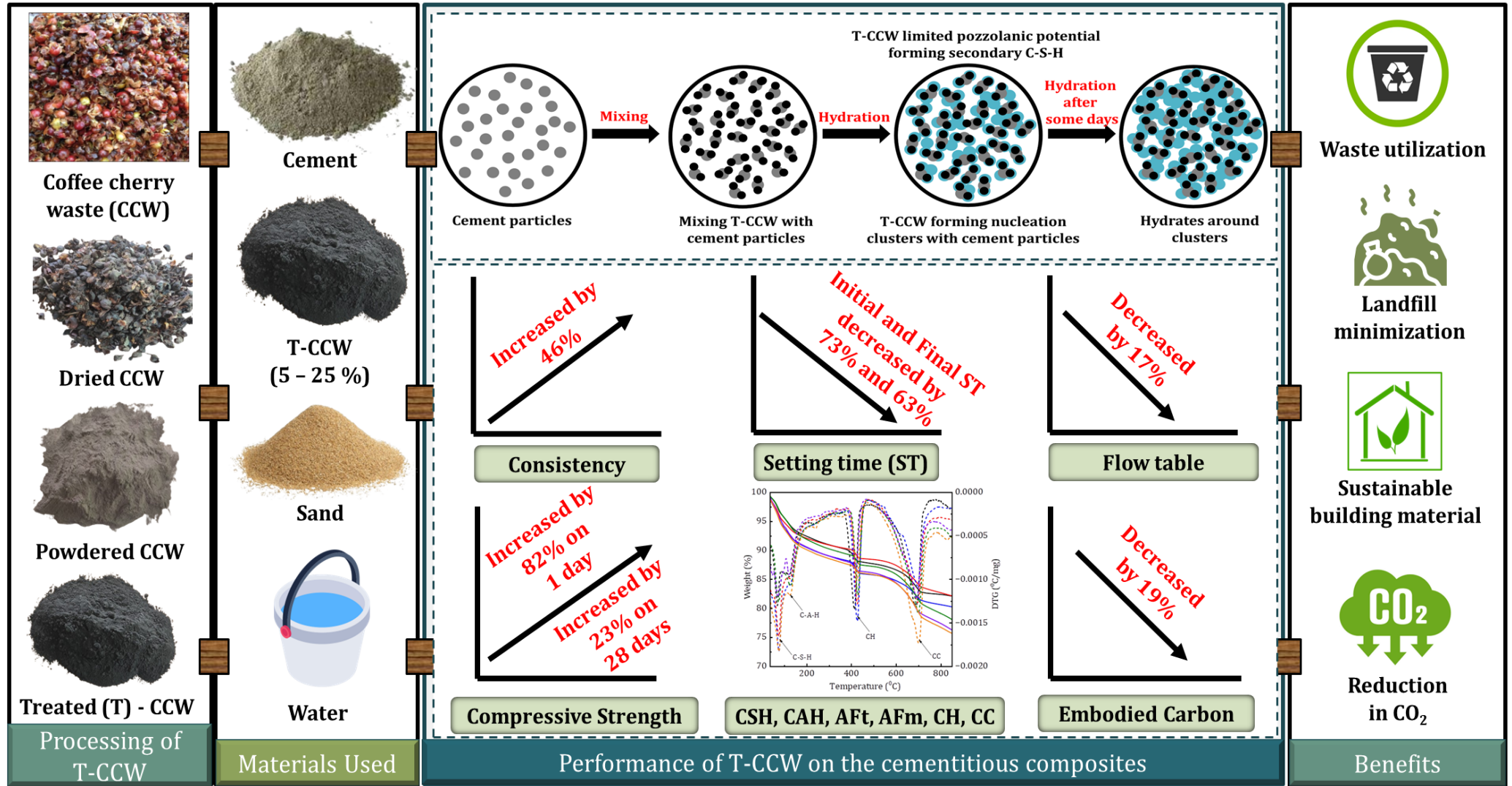
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14 **Highlights**

- 15 • Coffee cherry waste (CCW), a byproduct of coffee processing, was treated and investigated
16 as a potential sustainable material to partially replace cement.
- 17 • Incorporating treated CCW up to 15% replacement of cement improved the compressive
18 strength of mortar at all curing ages.
- 19 • Hydration studies showed that treated CCW enhanced cement hydration and led to a denser
20 microstructure.
- 21 • Sustainability assessment revealed that incorporating treated CCW significantly improved
22 the environmental performance.
- 23 • A fishbone diagram was developed to identify the potential challenges in the development
24 and application of treated CCW in cementitious composites.

Graphical Abstract



26

Abstract

27 This study explores the potential of treated coffee cherry waste (T-CCW) as a partial
28 replacement of cement in mortar. T-CCW was characterized and incorporated into pastes and
29 mortars at 5% to 25% cement replacement. The main objectives were to examine the fresh and
30 hardened properties, hydration, and environmental assessment. Results showed that the high
31 specific surface area and porous structure of T-CCW particles increased water demand and
32 accelerated setting times. T-CCW incorporation of up to 15% enhanced compressive strength
33 at all curing ages due to improved hydration and limited pozzolanic reactions. Ultrasonic pulse
34 velocity indicated good homogeneity and compactness in T-CCW blended mortars.
35 Microstructural analysis revealed that T-CCW enhanced cement hydration, leading to a denser
36 matrix. Environmental analysis showed a reduced embodied carbon and cement intensity index
37 compared to the control mix. Overall, the optimal performance was observed at 15% T-CCW
38 replacement, significantly improving engineering properties and environmental impact.
39 Further, the fishbone diagram addresses various factors to optimize the use of T-CCW as a
40 cementitious composite. These findings demonstrate the potential of T-CCW as a sustainable
41 construction material, offering a promising pathway towards environmentally friendly and
42 resource-efficient building practices while addressing waste management in the coffee
43 industry.

44 **Keywords:** Treated coffee cherry waste, Waste management, Strength, Hydration,
45 Sustainability.

46 1. Introduction

47 The excessive production of anthropogenic carbon dioxide (CO₂) emissions seriously threatens
48 ecosystems, and the damage caused by these emissions is immutable [1]. Anthropogenic
49 activities that lead to this problem involve various processes, such as the use of fossil fuels for
50 energy generation and the production of ordinary Portland cement (OPC) [2]. The production
51 of OPC is responsible for approximately 7-8% of worldwide greenhouse gas emissions, making
52 it one of the construction industry's most environmentally impactful materials [3]. Various
53 strategies have been employed to reduce CO₂ emissions from cement production: a) improving
54 energy efficiency, b) substituting fuels, c) implementing carbon capture and storage
55 technologies, d) promoting effective cement usage, and e) using supplementary cementitious
56 materials (SCMs) [4, 5]. The improvement in cement production through certain strategies may
57 be limited due to the need for specialized equipment requirements, leading to higher costs [6,

58 7]. However, using SCMs as a replacement for OPC provides a viable alternative due to the
59 economic benefits of using locally sourced materials and the potential for large-scale
60 production. Industrial and agricultural wastes are feasible alternatives for cement binders.

61 Agricultural waste ashes have been extensively studied as SCMs to enhance the
62 sustainability and performance of cement-based materials. Various sources, including walnut
63 shell ash (WSA) [8], oat husk ash (OHA) [9], pistachio shell ash (PSA) [10], banana leaf ash
64 (BLA) [11], wheat straw ash (WSA) [12], barley straw ash (BSA) [13], palm oil fuel ash
65 (POFA) [14], rice straw ash (RSA) [15], bagasse ash (BA) [16], rice husk ash (RHA) [17], corn
66 cob ash (CCA) [18], corn stover ash (CSA) [19], etc., have shown promising results. These
67 materials exhibit diverse effects on cement properties. For instance, WSA improves pozzolanic
68 activity, setting time, and long-term strength by promoting fibrous calcium silicate hydrate (C-
69 S-H). OHA, calcined at 600 °C, can replace up to 20% of OPC without compromising strength.
70 PSA enhances workability, setting time, and at 10% replacement, yielding a 17% increase in
71 strength. BLA and WSA contribute to increased strength, with BLA reducing water absorption
72 and WSA enhancing the microstructure of cement concrete. BSA optimizes the pore structure
73 of concrete, leading to significant improvement in strength up to 15%. POFA in nano size can
74 improve the compressive strength up to 30% without any reduction in strength. RSA, BA, and
75 RHA contain significant amounts of silica, improving workability, strength, and durability.
76 Untreated CCA presents challenges due to its high potassium content, but pretreated CSA
77 enhances silica content, accelerating hydration and improving strength. The environmental
78 benefits of these SCMs are significant, reducing landfill waste, lowering CO₂ emissions, and
79 recycling agricultural waste into valuable construction material. However, each material's
80 performance depends on its composition, processing conditions, and percentage of cement
81 replacement, highlighting the need to optimize their application in cementitious composites.
82 While significant progress has been made in this research, many agricultural wastes remain
83 underutilized. This research explores coffee cherry waste (CCW) as a novel, sustainable
84 alternative for construction materials, contributing to the enduring efforts to develop more
85 environmentally friendly building practices.

86 The utilization of agricultural waste proves to be a cost-effective alternative compared
87 to industrial waste [20]. Despite its application in various industries, disposing of agricultural
88 waste through burning in open fields or using methods like stacking or landfilling can
89 negatively impact the environment. Improper waste management and by-products can lead to
90 soil degradation and water pollution. Also, landfill disposal leads to methane gas emissions,

91 which are significantly more potent than CO₂, regarding their impact on global warming [21].
92 Thus, recycling and reusing waste can benefit the environment, economy, and society [22].

93 Globally, coffee is considered a highly significant commodity, widely traded, and the
94 most consumed beverage [23]. The coffee industry significantly impacts the global economy,
95 with Brazil, Vietnam, Colombia, Indonesia, India, Ethiopia, and Honduras being the biggest
96 coffee producer countries, representing about 80% of global production [24]. It is concerning
97 to note that coffee consumption and waste generation have increased simultaneously. During
98 coffee production, a considerable amount of waste is generated, with 45% of the coffee
99 plantation left as residue [25]. Also, during the washing process, a large quantity of water
100 becomes contaminated and carries excessive carbon, which can harm the environment [26].
101 The technique used to process coffee can generate by-products throughout the various stages
102 of processing, including pre-roasting coffee, dry processing (coffee cherry waste), semi-dry
103 and wet processing (coffee pulp), and post-roasting coffee (coffee silverskin and spent coffee
104 grounds) [27]. Fig. 1 shows the steps involved in the coffee processing using the wet post-
105 harvesting method.

106 Coffee cherry waste (CCW) is the non-usable part of the coffee beans and is the by-
107 product primarily obtained through the wet method during coffee processing [28]. These wastes
108 cover the coffee beans for approximately 12% of the cherry's dry weight [29]. It has been
109 reported that for every tonne of coffee processed, the coffee processing industry produces
110 approximately 0.18 tonnes of CCW [30]. The coffee cherries are taken to a processing facility
111 where the under-ripe and overripe cherries are separated and treated accordingly. The seeds are
112 then extracted from the cherries using a de-pulping machine. Approximately 90% of the coffee
113 cherry waste is often discarded into a compost pile or dumped in landfills or water bodies. The
114 improper disposal of these coffee cherry waste poses significant environmental problems for
115 the soil and water around the farmland due to its high acidity and high levels of caffeine,
116 tannins, and other polyphenols [31]. These wastes can acidify the soil, making nutrients
117 unavailable for crops. Additionally, the long-term release of caffeine into water bodies
118 negatively impacts the aquatic environment. Further, decomposing coffee cherries releases
119 toxic mycotoxins that grow on coffee cherries, causing negative effects on the central nervous,
120 cardiovascular, and respiratory systems [32]. The steps involved in generating CCW are shown
121 in Fig. 2.

122 With an estimated increased production of coffee in the future, waste management
123 policies for CCW must be implemented for proper handling and application, contributing to a
124 circular and sustainable economy. Studies have investigated the possibility of using spent

125 coffee grounds, but significant research has yet to be done on utilizing CCW as a construction
126 material. This motivated the exploration of the potential of CCW as an alternative construction
127 material to address the significant environmental challenges associated with cement production
128 and the improper disposal of this byproduct.

129 According to the literature, more research is needed on the possible utilization of CCW
130 as a substitute for cement. It is clear from previous studies that processing methods like burning
131 and grinding significantly impact reactivity [33]. Hence, this research explores the possibility
132 of using treated coffee cherry waste (T-CCW) for normal-strength mortar application. The
133 following objectives were studied to investigate the feasibility of utilizing T-CCW as a
134 construction material.

- 135 i) To perform an in-depth material characterization of T-CCW.
- 136 ii) To investigate the effect of T-CCW on cement paste and mortar properties.
- 137 iii) To study the effect of T-CCW on the hydration of cement paste.

138 An extensive experimental program has been carried out to answer these objectives. Different
139 microstructural analyses are performed to investigate the characterization of T-CCW by a) X-
140 ray diffraction (XRD), b) Thermogravimetric analysis (TGA), and c) Scanning electron
141 microscopy (SEM), d) Energy dispersive spectroscopy (EDS) e) Fourier transform infrared
142 spectroscopy (FTIR). Modified Chapelle tests were performed to assess the pozzolanicity. For
143 the fresh properties of the T-CCW paste, tests such as a) consistency, b) setting time, and c)
144 mini-slump flow are considered. The study investigates the hardened properties of T-CCW
145 mortar. The hydration characteristics of T-CCW paste are conducted using SEM, XRD, TGA,
146 and FTIR. A fishbone diagram is drawn to identify, analyze, and display the potential causes of
147 using T-CCW in cement.

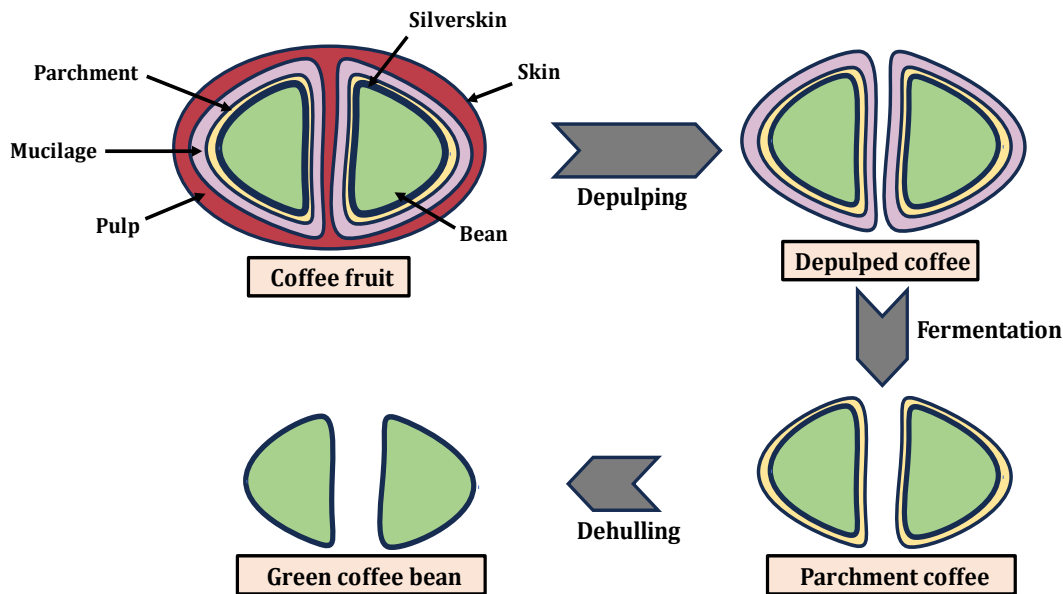


Fig. 1. Steps involved in the coffee processing using the wet post-harvesting method

148 **2. Materials and methods**

149 **2.1 Materials**

150 **2.1.1 Cement and Fine Aggregate**

151 The commercially available OPC of 53 grade conforming to IS 12269 is utilized in the present
 152 study. The chemical composition of OPC is shown in Table 1. The normal consistency is 29%,
 153 the specific gravity is 3.14, 160 min initial setting time and 310 min final setting time, and the
 154 specific surface area of OPC was 365 m²/kg. Natural river sand was used as a fine aggregate.
 155 The sand was washed, sundried, and sieved. The fine aggregate used in this study is categorized
 156 as zone 2, passing through a 2.36 mm sieve and retained on a 150 μm sieve. The water
 157 absorption, specific gravity, and fineness modulus were 0.95 %, 2.68, and 2.50, respectively.

158 **2.1.2 Treated coffee cherry waste**

159 The CCW used in the study is collected from the Chikkamagaluru district, Karnataka, India.
 160 Once coffee seeds are extracted, the leftover cherry is often discarded in landfills or water,
 161 leading to environmental issues like soil and water pollution. The CCW was collected from the
 162 disposal site, sun-dried for 7 days, and then dried in an oven at 105°C for 24 h to remove
 163 moisture. After cooling, the dried CCW was pulverized and grounded in a laboratory ball
 164 milled for 1 h at 60 rpm using stainless steel balls with a ball-to-powder ratio of 10:1. These
 165 specific durations and speeds were chosen to achieve the optimal particle size reduction. The
 166 SEM image of the grounded CCW is illustrated in Fig. S1. Later, the milled CCW was calcined

167 at 550°C for 30 min [34]. This temperature was selected based on the thermogravimetric
 168 analysis (TGA) results for the grounded CCW. The TGA curve of the grounded CCW is
 169 presented in Fig. S2. At this temperature, all the thermal decomposition processes have been
 170 completed, eliminating most of the organic matter, and the rate of weight loss has decreased,
 171 indicating a stable carbon structure. The duration of 30 min allows for complete thermal
 172 treatment and ensures uniform carbonization throughout the resulting material while
 173 minimizing the increase in the ash content. After cooling, the calcined CCW was sieved using
 174 a 75 µm sieve [35, 36]. This size was chosen to ensure uniformity in particle size distribution.
 175 Also, the reactivity of any material is often linked to its fineness. Finer particles have a larger
 176 surface area, which enhances their reactivity. The step-by-step procedure for T-CCW is
 177 illustrated in Fig. 3. The chemical composition of T-CCW is indicated in Table 1. The T-CCW
 178 exhibited a specific gravity of 2.12 and the Blaine’s specific surface area of 914 m²/kg. The
 179 size distribution of T-CCW particles is analyzed with a laser diffraction particle size analyzer,
 180 shown in Fig. 4. It confirms that the D10 (i.e., the diameter of the particles at 10% passing),
 181 D50, and D90 values of T-CCW are 3.03 µm, 19.21 µm, and 53.99 µm with mean particle size
 182 24.66 µm.

183 **Table 1** Chemical composition of OPC and T-CCW

Oxides (%)	OPC	T-CCW
SiO ₂	19.15	28.65
Al ₂ O ₃	5.67	10.30
Fe ₂ O ₃	4.87	10.72
CaO	63.25	11.77
MgO	1.56	1.8
SO ₃	2.26	2.64
Na ₂ O	0.35	0.22
K ₂ O	0.53	11.88
P ₂ O ₅	0.11	3.01
LOI	1.13	18.75

184 **2.2 Mix proportions**

185 Five mixes of T-CCW were prepared and investigated to study the impact of T-CCW on pastes
 186 and mortars. The mix proportions of pastes and mortars used in the study are given in Table 2.
 187 In this study, the water-to-binder ratio of 0.48 and sand-to-binder ratio of 2.75 were selected

188 based on the trial mixes to achieve a workable consistency and optimal performance of mortar
189 mixes. [20]. The variation of T-CCW replacement from 5% to 25% at 5% intervals was chosen
190 to systematically investigate the effect of incremental additions on the fresh and hardened
191 properties, providing a comprehensive understanding of the material's potential as a sustainable
192 cement replacement. The reference is labelled as C0 and the other C5, C10, C15, C20, and C25,
193 respectively. Based on the mixture proportion, the fresh and hardened characteristics of pastes
194 and mortar are conducted.

195 **2.3 Preparation of mixtures**

196 Cement paste was prepared using a high-speed stirrer, IKA RW 20, with a max speed of 2000
197 rpm. In the first stage, dry materials are mixed at 200 rpm for 60 s. Water is added to the mixture
198 and mixed for 60 s at 500 rpm. The paste is left to rest for 90 s in the next stage before being
199 continuously mixed for 90 s at 1500 rpm in the final step. Mortar mixes were prepared using a
200 standard Digi mortar mixer. Initially, all materials (cement, sand, T-CCW, and water) are
201 manually mixed. The materials are mixed at low speed for 60 s, then 30 s at high speed. The
202 mortar mixer rests for 90 s to clean the sides of the bowl before being mixed for another 60 s
203 at high speed [20]. Each mortar mix was kept at a flow value of 110 ± 5 mm or higher. The
204 mortar mixture was poured into cube-shaped moulds measuring 50 mm and 70.7 mm,
205 respectively. The moulds were vibrated to remove entrapped air, filled with the mixture, and
206 left for 24 h. They were then placed in a curing tank for water curing. Various tests were
207 conducted at the respective ages.

208 **2.4 Testing of samples**

209 The flowchart of the experimental program is illustrated in Fig. 5.

210 **2.4.1 Characterization of T-CCW**

211 X-ray fluorescence (XRF) was used to analyze the sample's chemical composition. XRD is a
212 method to examine the crystallographic structure of materials. The microstructure of the sample
213 was analyzed using SEM. TGA is a thermal analysis technique used to study changes in sample
214 weight as temperature increases under controlled conditions. FTIR is an analytical method that
215 uses infrared absorption and emission spectra to identify and analyze the chemical composition
216 of substances.

217 **2.4.2 Tests on pastes, fresh and hardened properties of mortars**

218 The Vicat apparatus was used to determine the standard consistency and setting times of cement
219 paste samples for both control and T-CCW, following IS 4031- Part IV and V. The mini-slump

220 cone test measured the flow of fresh cement paste by lifting the filled cone and measuring the
 221 increase in its diameter in perpendicular directions [37]. All the mortar mixes were prepared
 222 using a Digi mortar mixer according to IS 2250-1981. The mixer was used to mix fine
 223 aggregate, Portland cement, T-CCW, and water. After mixing, the flow table test was conducted
 224 in accordance with IS 4031 (Part 7)-1988. The mortar was placed on the table, and the flow
 225 was recorded by measuring the spread diameter of each mixture after 25 drops. The
 226 compressive strength of mortar was measured using 50 mm cubes that were water-cured for 1,
 227 3, 7, and 28 days. Vertical axial loading was applied, and the maximum load was recorded to
 228 calculate the compressive strength. Ultrasonic pulse velocity (UPV) is a non-destructive testing
 229 method that measures pulse velocity to evaluate properties such as homogeneity, cracks, and
 230 voids in hardened mortar. The mortar cubes, measuring 70.7 mm, are tested for UPV using a
 231 PUNDIT portable device.

232 **2.4.3 Hydration Studies of T-CCW**

233 The pastes were moulded and stored in lime water to prevent leaching. XRD, TGA, FTIR, and
 234 SEM studies were performed on 7-day hydrated cement paste samples. The samples were
 235 treated with isopropanol and diethyl ether to stop the hydration process and stored free from
 236 carbonation until tested. Samples were taken from the interior of the crushed specimens for
 237 SEM analysis. The specimens were ground and sieved through a 75 µm sieve for XRD, TGA,
 238 and FTIR. For TGA, the weight loss was then monitored from 30 to 900 °C, with a heating rate
 239 of 10 °C/min. The amount of calcium hydroxide or portlandite (CH) present in cement samples
 240 can be determined using Eq. (1) [11].

$$241 \quad \text{CH} = \frac{(W_{400} - W_{500})}{W_{500}} \cdot \frac{74}{18} \cdot 100 \quad (1)$$

242 Based on previous studies, bound water (BW) was determined using Eq. (2) [11]. The higher
 243 BW indicates a more significant formation of hydrates [38, 39].

$$244 \quad \text{BW} = \frac{(W_{40} - W_{500})}{W_{500}} \cdot 100 \quad (2)$$

245 W40, W400 and W500 – Weight loss at 40 °C, 400 °C and 500 °C. The breakdown of hydrates
 246 can be separated into three main stages. The initial stage, below 400 °C, includes the
 247 disintegration of hydrates. The second stage, between 400 to 500 °C, is mainly dominated by
 248 the dehydroxylation of CH. The third stage is characterized by the decarbonization of calcite
 249 (CC) [40]. For FTIR, pellets were created with 1 mg sample and 100 mg KBr, with spectra
 recorded in the wavenumber range of 4000–400 cm⁻¹. The crystalline phases present in the
 samples were evaluated through XRD in the 2θ range of 10 – 80° for samples.

250 **2.4.5 Environmental analysis and fishbone diagram**

251 Sustainability assessment and the cement intensity index (CII) were calculated for the mortar
252 mixes and the hardened properties of the mortar, followed by a fishbone diagram as an
253 analytical tool examining the potential challenges associated with the use of T-CCW.



Fig. 2. Steps involved in generating CCW: 1) Coffee fruit 2) Harvesting of the coffee fruit 3) Washing coffee fruit 4) Depulping coffee fruit 5) Cherry waste dumped as landfill 6) Cherry waste dumped in the water bodies

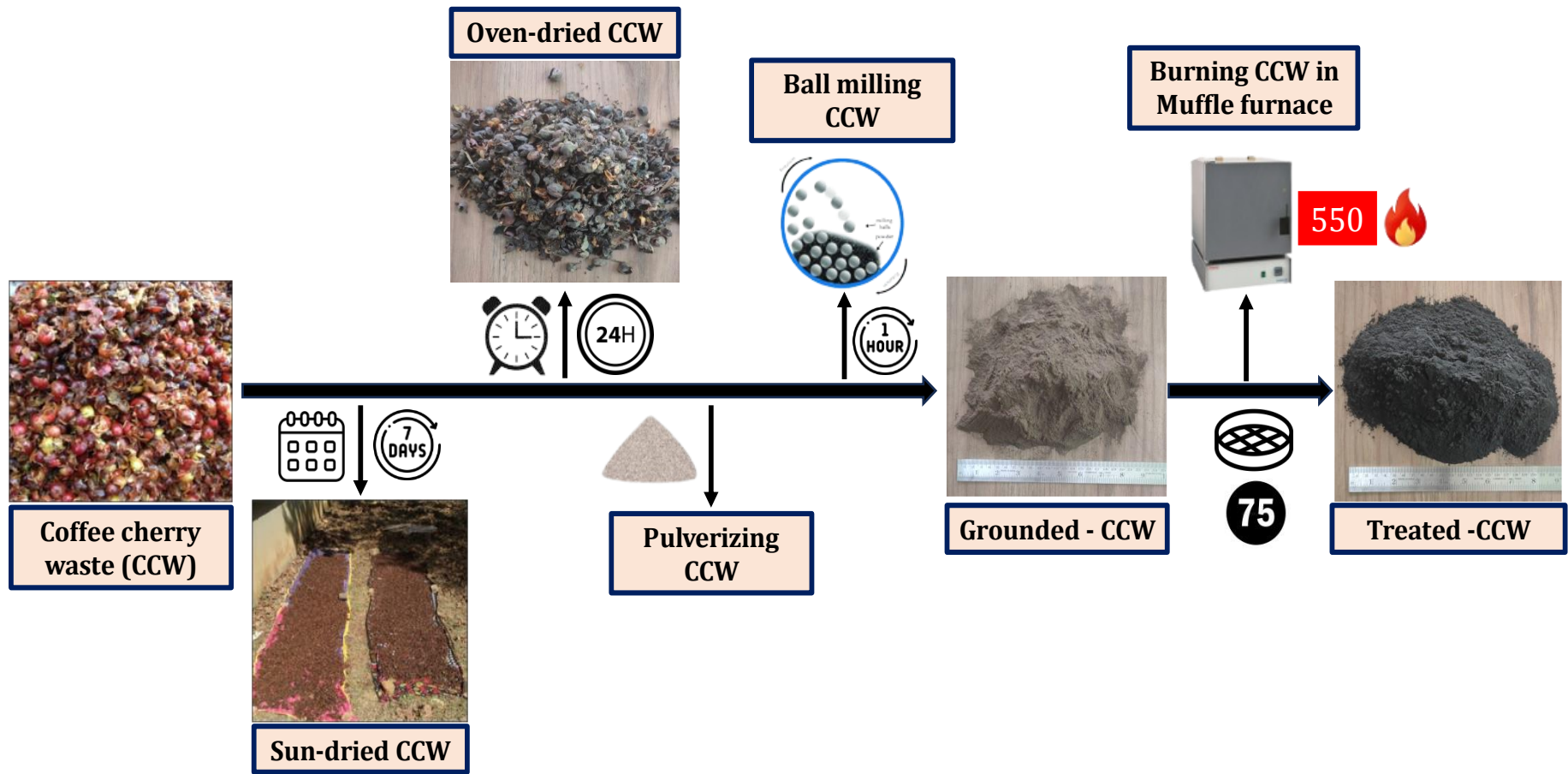


Fig. 3. Step-by-step procedure for the preparation of T-CCW

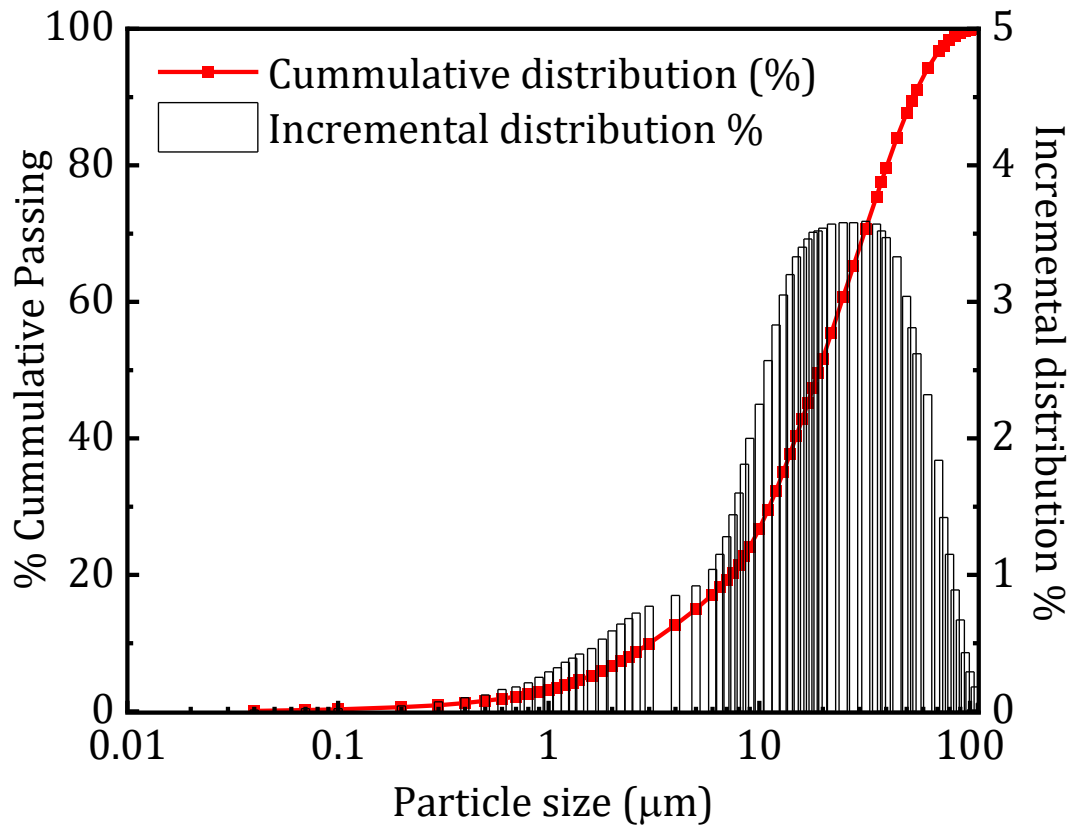


Fig. 4. Particle size analysis of the T-CCW

256

Table 2 Mix proportions of T-CCW – OPC based blended paste and mortar

Mixes	OPC (g)	T-CCW (g)	Water (g)	Sand (g)
C0	450.0	0.0	216	1237.5
C5	427.5	22.5		
C10	405.0	45.0		
C15	382.5	67.5		
C20	360.0	90.0		
C25	337.5	112.5		

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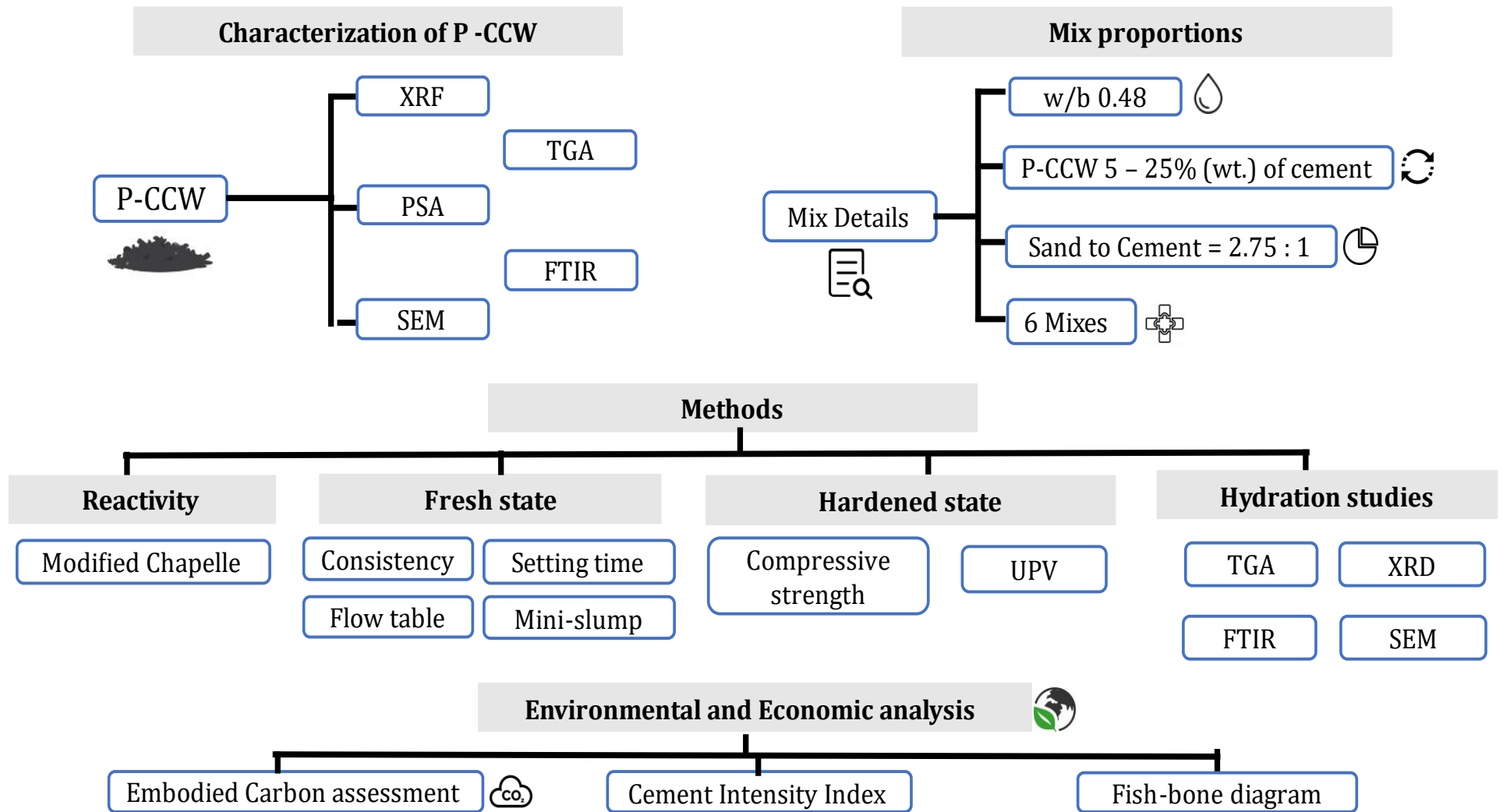


Fig. 5. Flowchart of the experimental program

258 **3. Results and discussions**

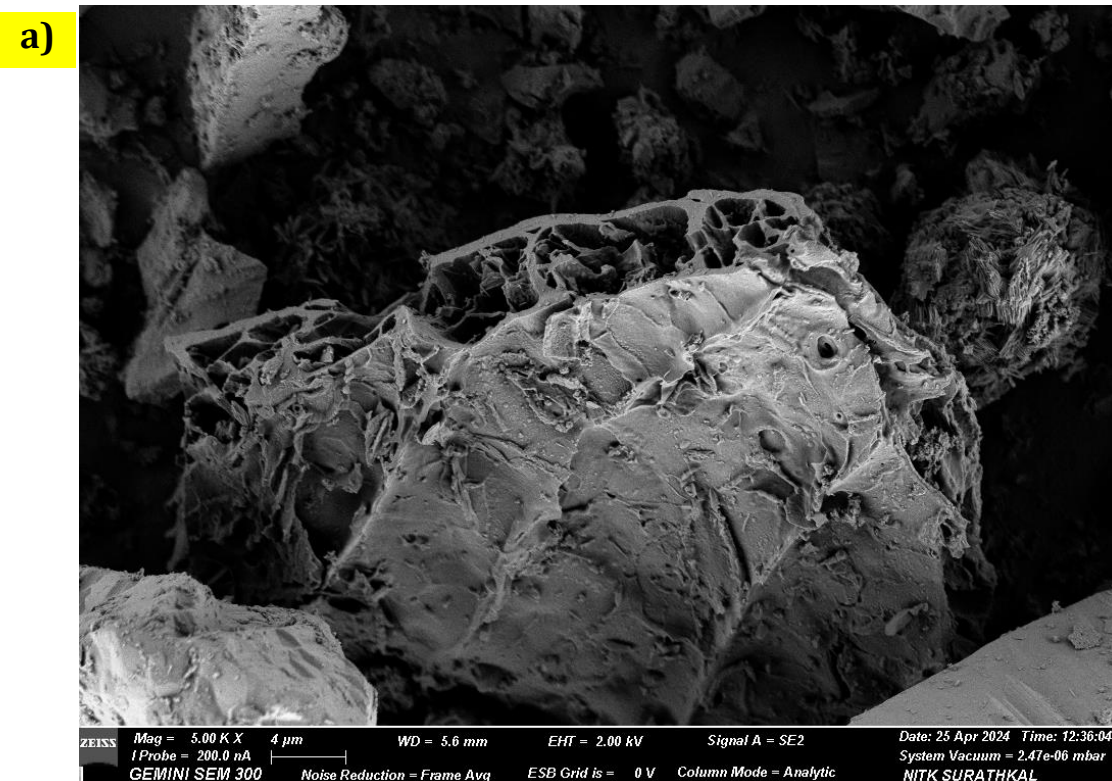
259 **3.1 Characterization of T-CCW**

260 The morphology of the T-CCW is studied by using SEM, as shown in Fig. 6(a). T-CCW
261 particles exhibit irregular shapes with wrinkled and rugged surfaces [31, 41]. Further, T-CCW
262 particles exhibit the presence of numerous pores on the surface, indicating the release of
263 volatile and organic matter from the untreated CCW during the thermal decomposition process.
264 These pores can absorb water and serve as an internal curing agent. The SEM of the untreated
265 CCW is illustrated in Fig S1. The EDS analysis of the T-CCW is presented in Fig. 6(b).

266 The FTIR spectra of the T-CCW sample exhibit several peaks, indicating the presence
267 of various functional groups. The broad peak at 3300 cm^{-1} is attributed to the stretching
268 vibrations of O-H groups [42], suggesting the presence of phenolic hydroxyl groups in the T-
269 CCW sample. The peak at 2986 cm^{-1} is assigned to the C-H stretching vibration of methyl
270 groups present in organic compounds such as proteins, lignin, and cellulose. These peaks may
271 indicate the presence of caffeine [43]. The absorption band at 1420 cm^{-1} corresponds to the
272 bending vibration of C-H bonds in methyl groups [44]. The peak at 1020 cm^{-1} is associated
273 with the stretching vibration of C-O groups in the C-O-H glycosidic bonds, which are
274 characteristic of galactomannans, a class of polysaccharides present in the CCW [45]. The
275 peak at 874 cm^{-1} is attributed to the stretching vibrations of the C-O-C glycosidic bonds, further
276 confirming the presence of polysaccharides in the sample [46]. The band at 764 cm^{-1} indicates
277 the presence of aromatic C-H deformation vibration [47], suggesting the presence of aromatic
278 compounds derived from lignin and other aromatic components in the CCW.

279 TGA was employed to investigate the thermal decomposition characteristics of T-
280 CCW. The TG-DTG curves in Fig. 7(b) show three distinct thermal degradation zones. The
281 first zone, ranging from 30 to 190 °C with a peak at 55 °C, corresponds to the evaporation of
282 surface-bound water [31], resulting in approximately 3% weight loss. The second zone, ranging
283 from 190 to 500 °C, represents the decomposition of major organic compounds in T-CCW.
284 While no prominent peaks were observed in this region, a peak at around 440 °C suggests
285 significant cellulose content and the beginning of the lignin decomposition. Typically,
286 hemicellulose decomposition occurs between 200 to 300 °C, cellulose between 300 to 400 °C,
287 and lignin decomposition initiates at approximately 400 °C, extending to higher temperatures
288 [48]. The third zone, ranging from 500 – 900 °C with a peak at 660 °C, is attributed to the
289 endothermic degradation of lignin [49] and the breakdown of thermally stable organic

290 compounds [50]. The shift of this peak towards higher temperatures indicates enhanced thermal
 291 stability of the material at elevated temperatures.



Element	Wt. %	At %
C K	43.19	56.80
O K	29.18	28.81
Al K	1.50	0.88
Si K	18.63	10.48
K K	7.49	3.03

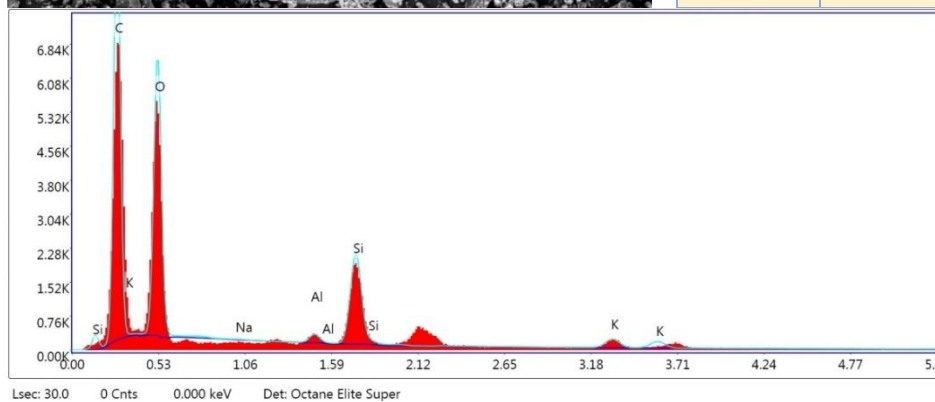


Fig. 6. a) SEM b) EDS of the T-CCW

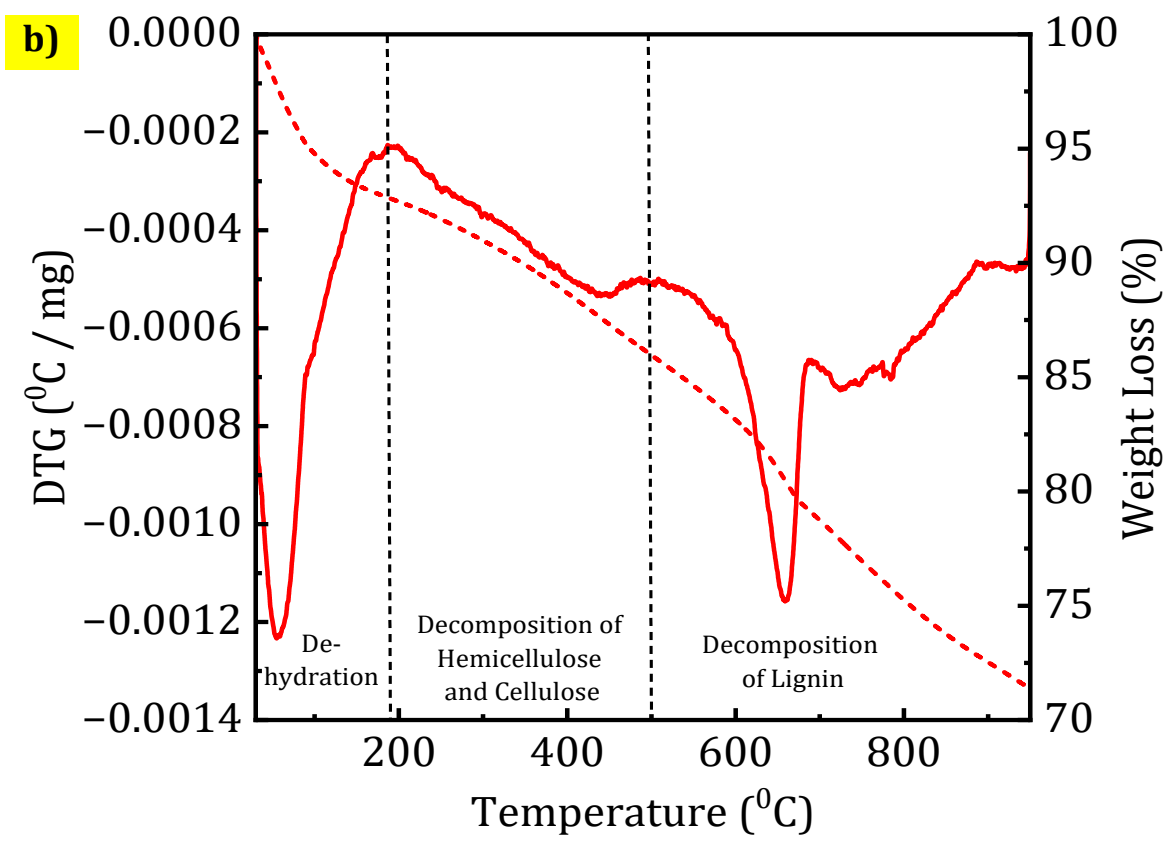
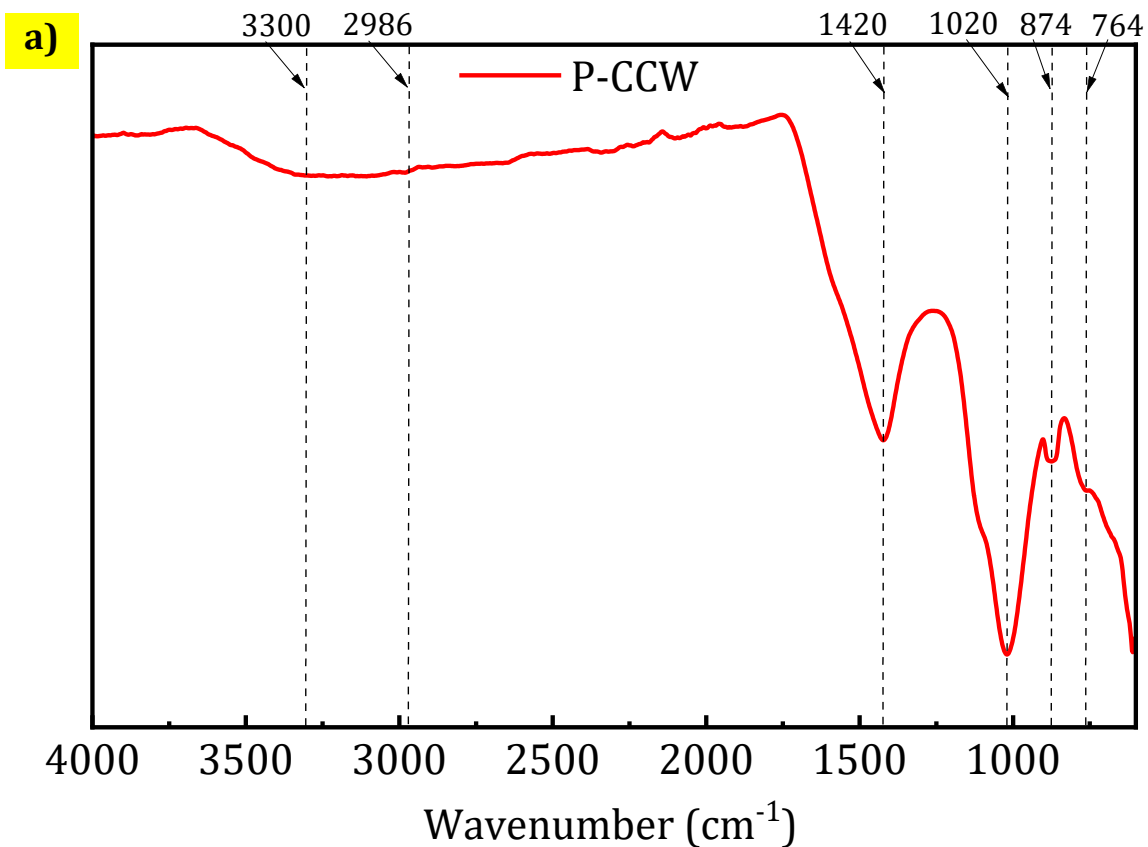


Fig. 7. a) FTIR b) TGA of T-CCW

293 **3.2 Modified Chapelle test to evaluate the reactivity of T-CCW**

294 The Chapelle activity test is a chemical method to measure the fixed quantity of calcium
295 hydroxide on pozzolanic materials to define their pozzolanic activity. The procedure to conduct
296 the modified Chapelle test is as per NF P 18-513 standards [51]. Fig illustrates the steps carried
297 out for the Chapelle test of T-CCW. Using the formula given in eqn

$$\text{Amount of Ca (OH)}_2 \text{ fixed} = 2 \times ((V_1 - V_2) / V_1) \times (74/56) \times 1000$$

298
299 Where V_1 is the volume of 0.1N HCl obtained by blank, which is equal to 9.4, and V_2 is the
300 volume of 0.1N HCl obtained by the reaction of the T-CCW, which is equal to 9.0. The
301 Ca (OH)_2 fixed for T-CCW is 112 mg of $\text{Ca (OH)}_2/\text{g}$. This indicates that T-CCW consumes less
302 CaO than traditional SCMs like fly ash, which typically shows around 666 mg of $\text{Ca (OH)}_2/\text{g}$
303 [52]. Based on the results of the Chapelle test, T-CCW cannot be classified as a traditional SCM
304 due to its significantly lower CaO consumption compared to conventional SCMs like fly ash,
305 slag, or silica fume. However, T-CCW demonstrates some pozzolanic properties that contribute
306 to cementitious mixtures. Therefore, T-CCW can be proposed as an alternative sustainable
307 material with unique characteristics that exhibit lower pozzolanic potential and can act as a
308 filler material.

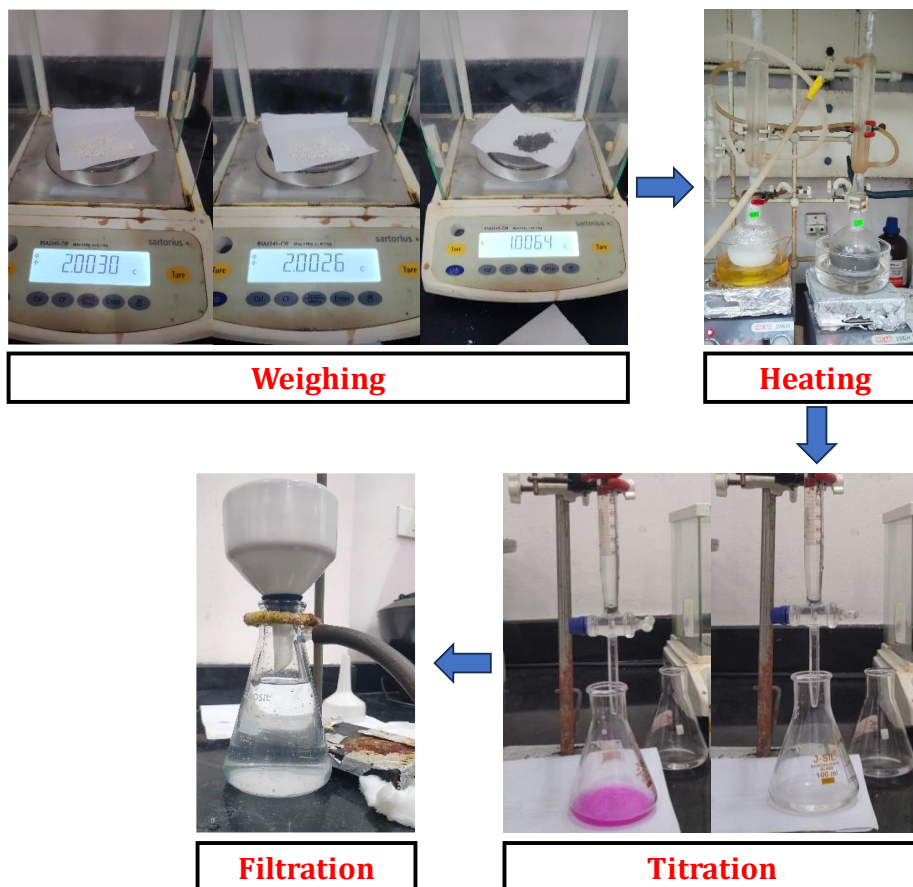


Fig. 8. Experimental setup for Modified Chapelle test.

309 **3.3 Normal Consistency**

310 The consistency of the blended cement pastes incorporating T-CCW was determined by the
311 percentage of water demand, according to the replacement levels of T-CCW, as shown in
312 Fig. 9. The control mix exhibited a consistency of 28.5%. As the T-CCW replacement levels
313 increased, the water demand for achieving consistency progressively increased to 29%, 31.5%,
314 33%, 38.5%, and 41.5%, respectively. The water requirement for the consistency of the blended
315 cement pastes increased up to 46% compared to the control mix. This significant increase in
316 the water demand for T-CCW blended cement pastes can be attributed to the porous nature of
317 the T-CCW particles, as evidenced by the SEM images in Fig. 6(a). The porous structure of T-
318 CCW particles allows them to retain water within their pores, effectively reducing the free
319 water available and necessitating additional water to maintain the desired consistency [53]. The
320 results of using other biomass ashes have also increased the water demand for the blended
321 cement pastes [10, 20].

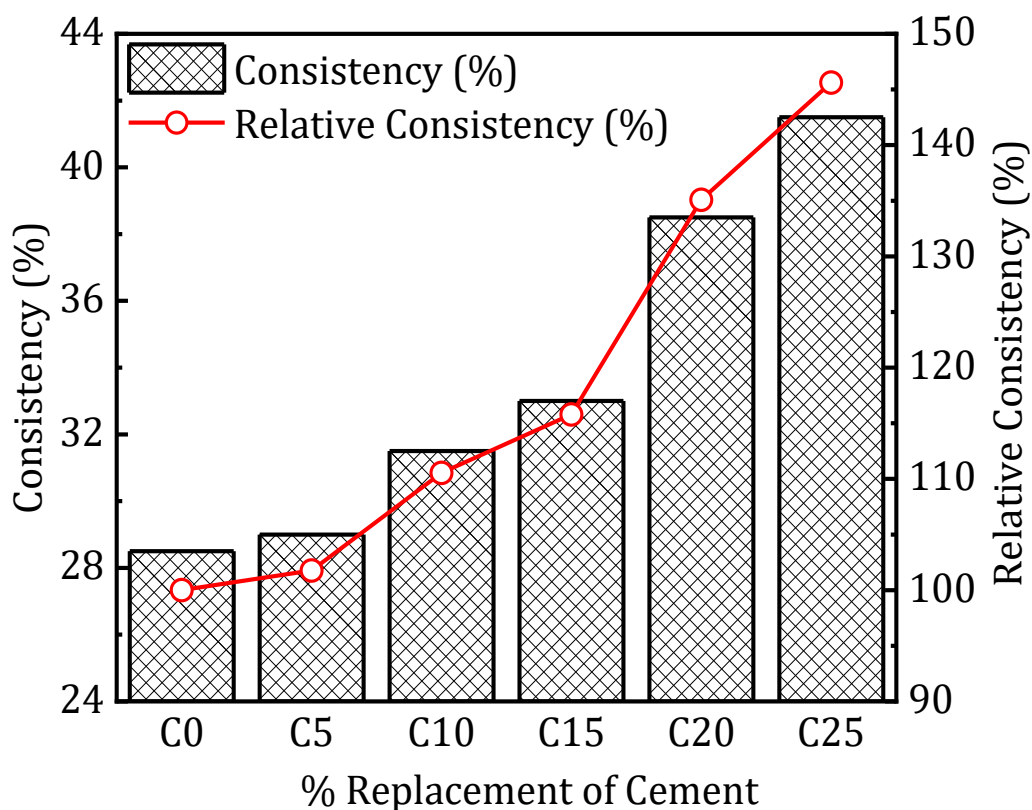


Fig. 9. Consistency of the blended cement

322 **3.4 Setting time**

323 The hydration process is responsible for the setting of the cement paste. The initial and final
324 setting times for various percentages of T-CCW were measured using Vicat's apparatus, and
325 the results are shown in Fig. 10. The experimental results showed that the increase in the

326 percentage of T-CCW leads to a substantial reduction in the initial and final setting times of the
 327 blended cement pastes. The addition of 25% T-CCW resulted in a 73% and 63% decrease in
 328 the initial and final setting times compared to the control cement paste. This behaviour can be
 329 attributed to three mechanisms. (i) The high surface area and porous structure of T-CCW
 330 material provide nucleation sites for the formation of hydration products, and the water
 331 retention capacity of the T-CCW within the pore structure ensures a continuous supply of water,
 332 maintaining the hydration reactions and contributes to faster setting times [54]. (ii) The higher
 333 alkali content in T-CCW has accelerated the hydration process, thereby affecting the setting
 334 time of the blended cement paste [55].

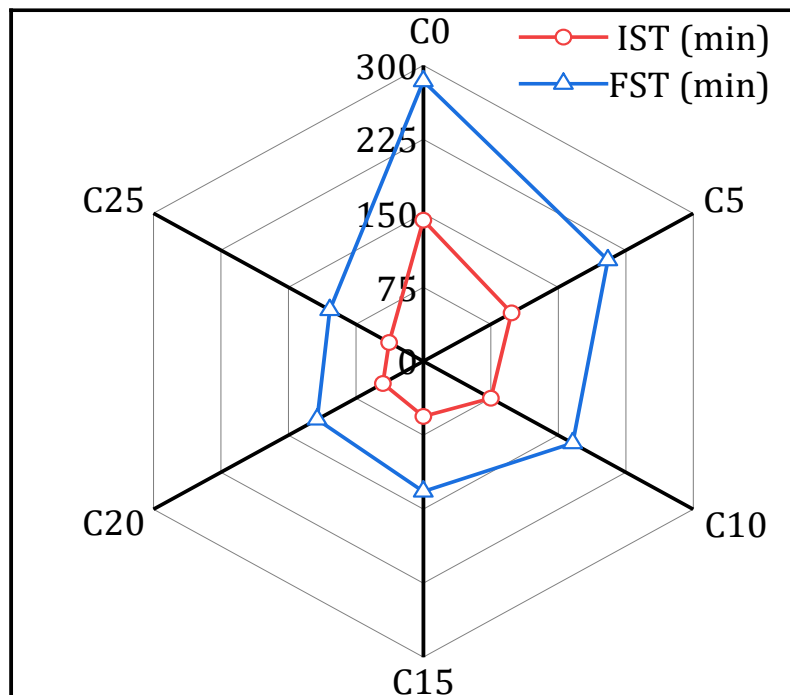


Fig. 10. Setting time of the blended cement

335 3.5 Mini slump flow and Flow table

336 The relationship between the percentage of T-CCW content and mini-slump flow is illustrated
 337 in Fig. 11. The mini-slump diameter of the control paste was 142 mm at 0 min and decreased
 338 to 110 mm at 25 min. However, for the blended cement pastes, the results indicate that the
 339 increase in the percentage of T-CCW content reduced the mini-slump flow diameter at all the
 340 time duration. The significant water absorption, high specific surface area, and porous structure
 341 of T-CCW are attributed to the decrease in flowability [53].

342 Furthermore, a flow table test conducted on the mortar with varying percentages of T-
 343 CCW revealed that the flow table values decreased with the increase in T-CCW content, as
 344 presented in Table 3. The flow table value of the blended cement mortar decreased by 5%, 8%,

345 10%, 12%, and 17%, respectively, compared to the control mortar. The incorporation of T-
 346 CCW necessitates additional water due to its irregular geometry and porous nature, as observed
 347 by SEM, leading to lower flow values [56].

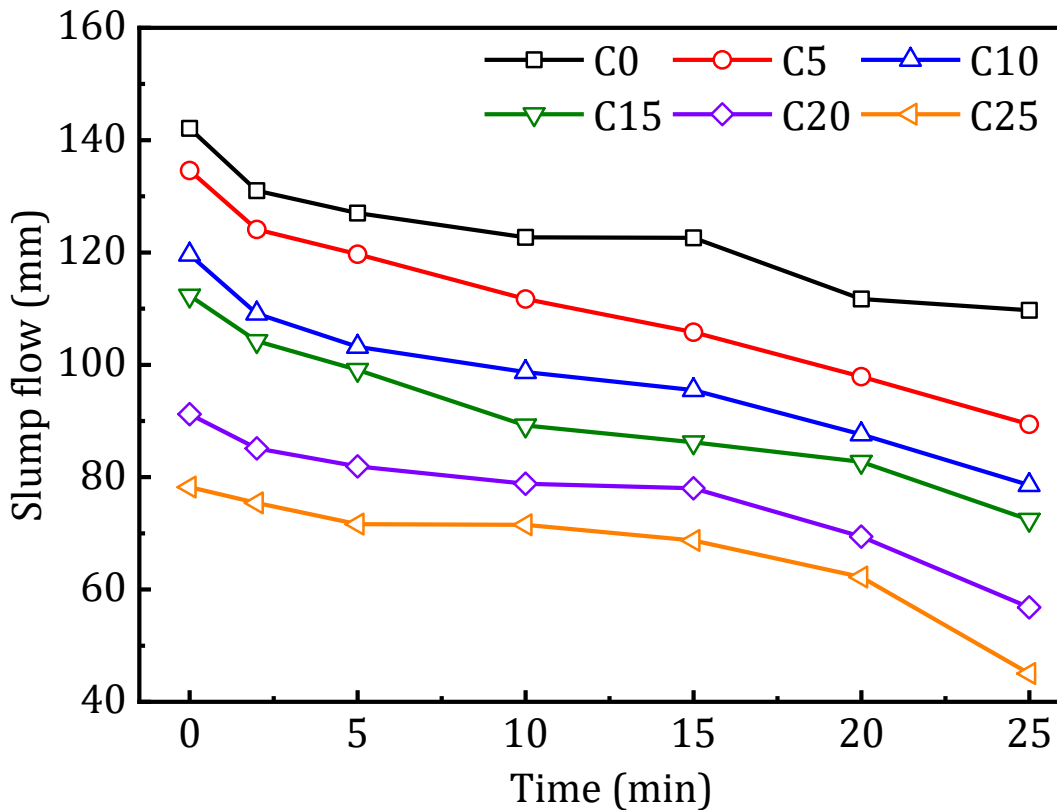


Fig. 11. Mini slump flow and flow table value of the blended pastes

348

Table 3 Flow table of the blended cement mortar

Mix	Flow table value (mm)
C0	155
C5	148
C10	143
C15	140
C20	136
C25	129

349 3.6 Compressive Strength

350 One of the critical parameters governing the structural viability of mortar mixes is compressive
 351 strength, which directly correlates to the load-carrying capacity. The compressive strength of
 352 the control and blended cement mortar samples incorporating T-CCW at various replacement
 353 levels was conducted at 1, 3, 7, and 28 curing days, as shown in Fig. 12. The influence of T-

354 CCW has a notable impact on the strength development of the mortar mixes. An increase in T-
355 CCW percentage up to 15% in the mortar increases compressive strength compared to the
356 control mortar at all the curing ages. The incorporation of 15% T-CCW yielded the highest
357 compressive strength values of 13.6, 24.1, 32.6, and 45.1 MPa, exhibited an enhancement of
358 82%, 53%, 49%, and 23% compared to the control mortar at 1, 3, 7, and 28 days of curing,
359 respectively. Firstly, the improved strength can be attributed to the larger surface area and
360 porous and hydrophilic nature of T-CCW, which facilitates water absorption during mixing and
361 subsequent release during self-desiccation, effectively completing the hydration process [57,
362 58]. Secondly, T-CCW favours the development of pozzolanic reactions, allowing secondary
363 calcium silicate hydrate (C-S-H) gel formation. Additionally, the high alkaline content of T-
364 CCW accelerates the hydration reaction, enhancing early strength in cementitious composites
365 [59]. Further increasing the T-CCW content beyond 15% reduced strength, potentially due to
366 the low-filling effect [60] and the increase in pores [61]. However, the compressive strength of
367 all the mortar mixes was within the standard value specified by ASTM C270 and ASTM C90.
368 Compared to T-CCW with other SCMs of similar mix proportion, 15% T-CCW blended
369 cementitious composite achieved a 28-day compressive strength of 45.1MPa, which is
370 comparable with the 20% fly ash replacement (42.7 MPa) reported by Siddique. [62]. However,
371 a 30% GGBS replacement studied by Oner and Akyuz (2007) yielded a slightly higher strength
372 of 47.5 MPa [63]. These comparisons highlight that T-CCW can achieve similar or better
373 strength performance at lower replacement levels compared to other common SCMs because
374 of its nucleation and limited pozzolanic effect.

375 In addition, a two-way ANOVA was employed to understand the influence of T-CCW
376 percentage and concrete age on the strength, and the values are presented in Table S2. It is
377 evident from the analysis that the T-CCW percentage ($F(5, 63) = 12.4, p < 0.0001$) and age of
378 concrete ($F(3, 63) = 277.67, p < 0.0001$) have a significant influence on strength values.

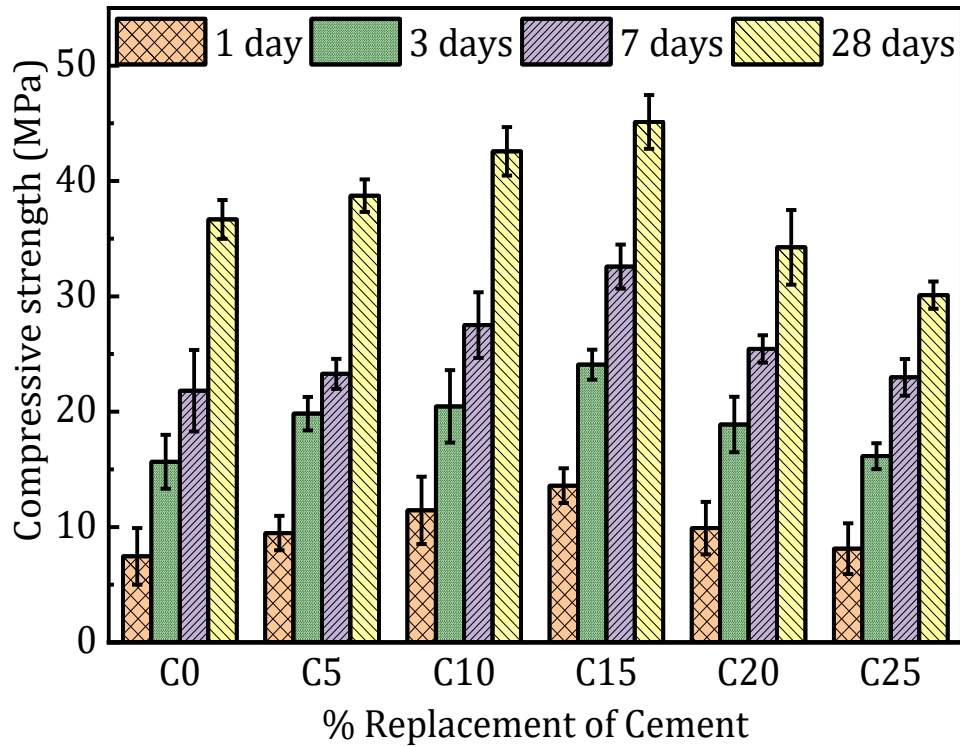


Fig. 12. Compressive strength of the blended cement

379 3.7 Ultrasonic pulse velocity

380 The UPV, a non-destructive method, was employed to evaluate the uniformity, homogeneity,
 381 and presence of voids or cracks in the hardened mortar specimens [20]. Fig. 13 illustrates the
 382 impact of T-CCW on the UPV values of the mortar at curing ages of 1, 3, 7, and 28 days. The
 383 UPV of the mortar ranged from 3258 to 3715 m/s on 1 day, 3459 to 4119 m/s on 3 days, 3724
 384 to 4296 m/s on 7 days, and 3978 to 4525 m/s on 28 days. A notable increase in the UPV values
 385 is observed with the increase in the curing period. The C15 mix, exhibiting the highest
 386 compressive strength, also demonstrated the maximum UPV value, indicating a good
 387 correlation between UPV and the compressive strength of the mixes. This correlation suggests
 388 a positive influence of T-CCW on the homogeneity and compactness of the mortar matrix. The
 389 utilization of T-CCW causes a micro-filling effect, resulting in the development of additional
 390 nucleation sites and the development of secondary pozzolanic reactions that contribute
 391 effectively to the densification of the matrix [54]. However, higher replacement levels of T-
 392 CCW resulted in a less dense matrix with increased porosity. All the mortar mixes exhibited
 393 UPV values exceeding 3900 m/s at 28 days, indicating the mix is durable [64], and potentially
 394 advantageous for structural applications. The two-way ANOVA analysis showed that the
 395 percentage of T-CCW and curing age significantly affect UPV in concrete. Changes in T-CCW

396 percentage had a strong effect ($F(5, 63) = 39.16, p < 0.0001$), meaning that even small changes
 397 in T-CCW amount can noticeably influence UPV. Curing time had an even more significant
 398 impact ($F(3, 63) = 133.61, p < 0.0001$). The overall model was very strong ($F(8, 63) = 74.58,$
 399 $p < 0.0001$).

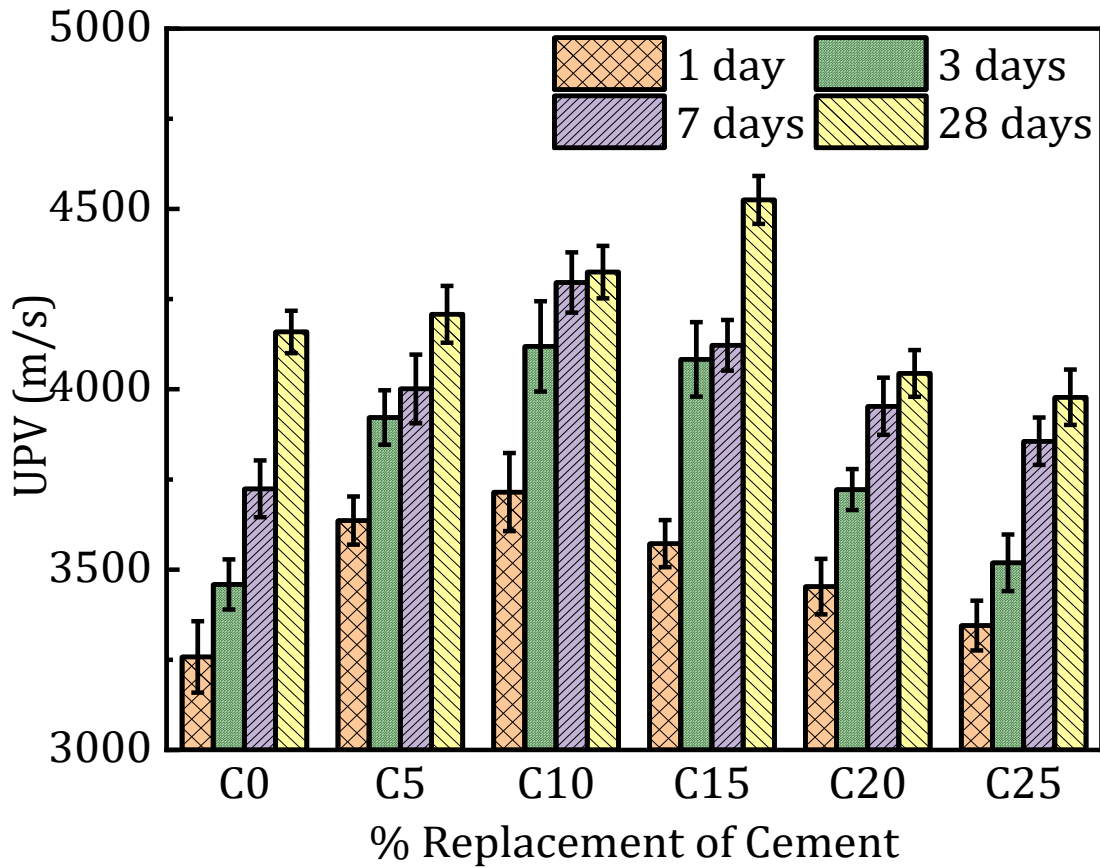


Fig. 13. UPV of the blended cement

400 4. Hydration studies

401 4.1 Scanning electron microscope

402 Fig. 14 presents the SEM images of C0, C10, and C25 pastes after 7 days. The SEM images
 403 show the formation of hydration products, such as the needle-shaped ettringite (AFt) crystals,
 404 distinctive hexagonal-prismatic crystals of portlandite (CH), and C-S-H gel. The T-CCW
 405 samples (C10 and C25) showed traces of monosulphate (AFm). The T-CCW admixed pastes
 406 exhibit a dense microstructure compared to control pastes, resulting in increased hydration
 407 products. The hydration process involves pore solution entering internal pores, facilitating the
 408 precipitation of hydration products within these pores [65], and the secondary pozzolanic
 409 reaction formed by the reaction of T-CCW with CH contributes to the strength development.
 410 Further, the fine particles of T-CCW act as nucleation sites, and the porous structure helps to

411 retain water, ensuring continued hydration and the formation of a denser matrix [66]. As a result
412 of these mechanisms, the SEM of the T-CCW blended pastes exhibits fewer voids compared to
413 the control.

414 **4.2 Thermogravimetric analysis**

415 TGA-DTG was used to investigate the influence of T-CCW on the formation of hydration and
416 carbonation products, as illustrated in Fig. 15(a). The three main peaks associated with the TGA
417 curves are dehydration of C-S-H, calcium aluminate hydrate (C-A-H), Aft, and AFm (105 –
418 400 °C), dehydroxylation of CH (400 – 500 °C), and decarbonation (500 – 900 °C) [67]. The
419 differential thermogravimetric (DTG) curve shows that the peak before 100 °C is attributed to
420 the enhanced hydration product formation due to T-CCW, which absorbs water from the
421 surrounding environment [65]. Also, the pozzolanic reactions facilitated the development of
422 secondary hydrates to occur. The second peak (400 – 500 °C) corresponds to the decomposition
423 of CH, which is directly related to the degree of cement hydration. The CH content directly
424 from the DTG curves cannot be interpreted correctly; therefore, the percentage of CH and BW
425 were quantitatively determined using Eq (1) and (2). The results of these calculations are
426 presented in Fig. 15(b) to provide a clear understanding of the hydration process. The increase
427 in T-CCW percentage resulted in a decrease in CH and an increase in BW. Despite the higher
428 w/c ratio, the reduction in CH is not only due to the dilution of the cement content but also the
429 secondary pozzolanic reaction between T-CCW and CH [11]. These findings agree with the
430 compressive strength results at 7 days. The third peak (600 – 750 °C) resulted from the
431 decomposition of CC, precipitated during the carbonation of C-S-H, and well-crystalline calcite
432 formed from the carbonation of portlandite.

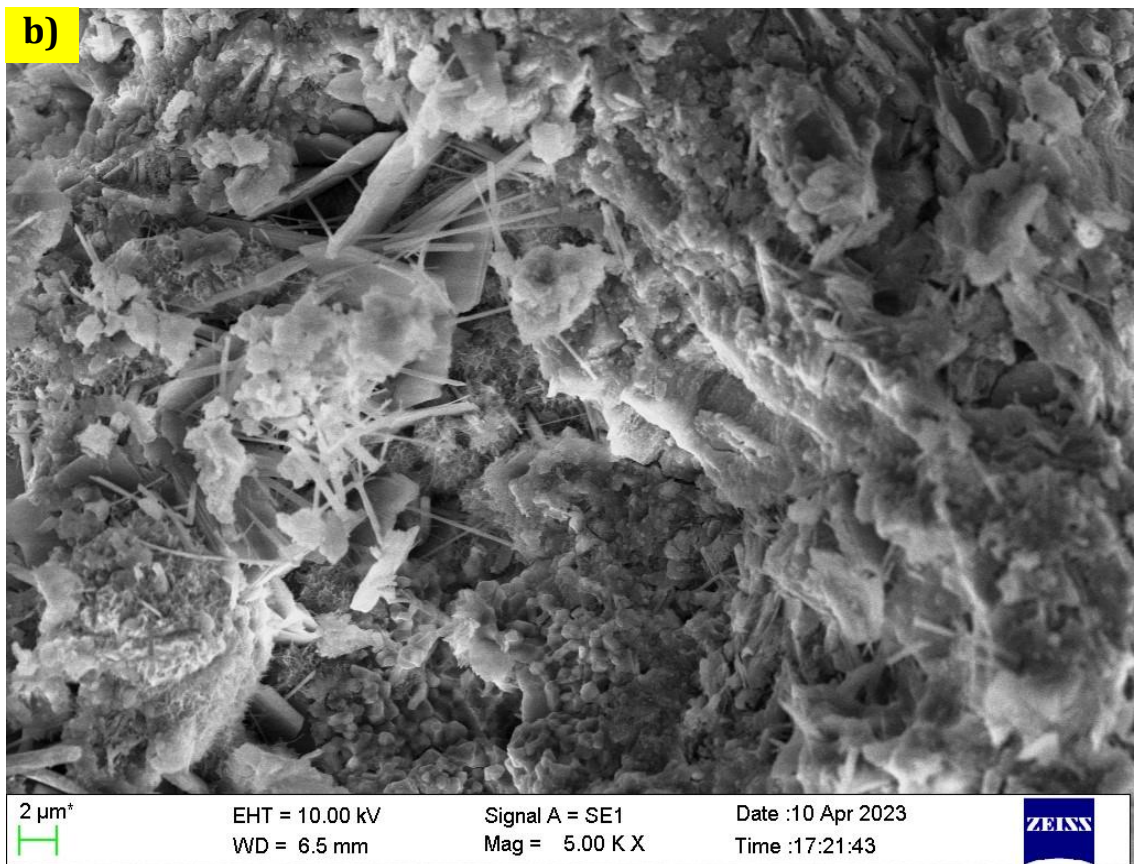
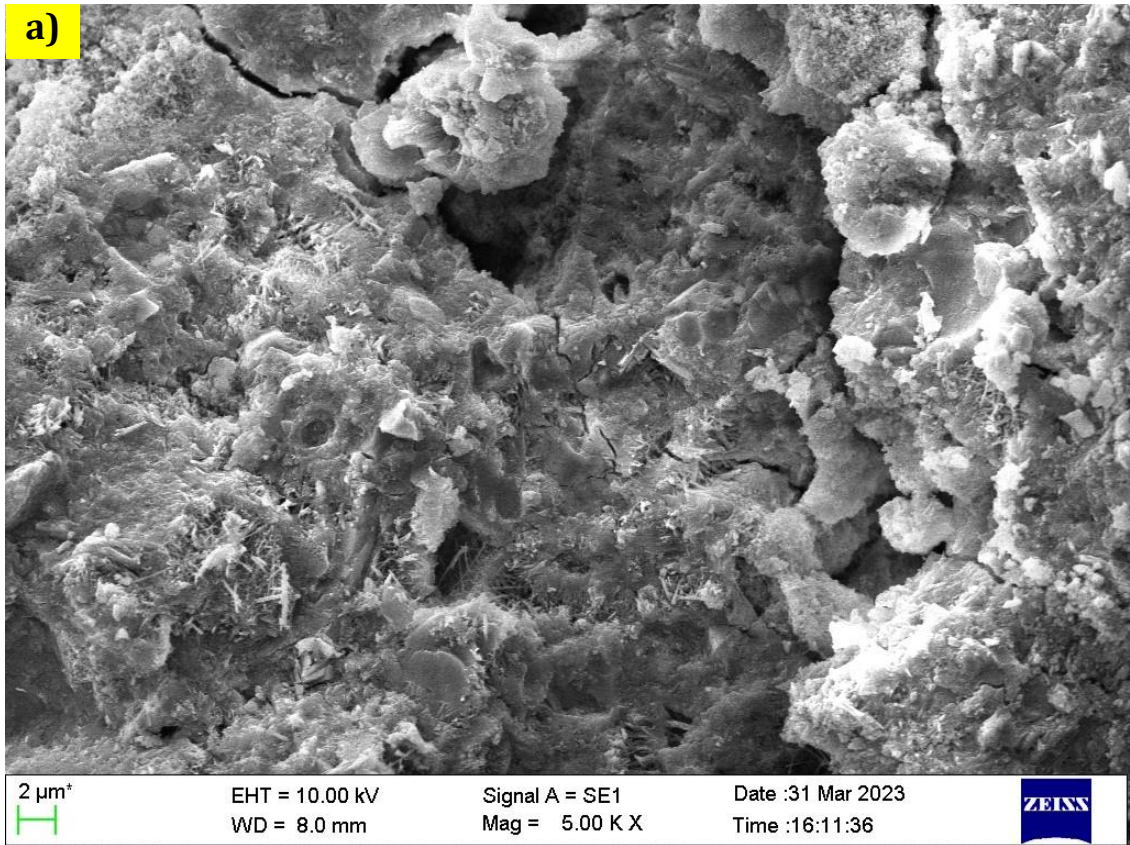
433 **4.3 X-ray diffraction**

434 The XRD patterns for the cement paste blended with T-CCW at 7 days are presented in Fig.
435 15(c). The results indicated that the incorporation of T-CCW enhanced the formation of
436 hydration products. The main crystalline phases detected in all the samples are CH (P), CC (C),
437 Tricalcium silicate (C₃S) (T), and Dicalcium silicate (C₂S) (D). The AFt and AFm phases are
438 observed in T-CCW blended samples with higher replacement levels. The characteristic peaks
439 of CH were identified at 18°, 34.08°, 47.08°, and 50.8°. The intensity of the portlandite peak
440 at 18° was higher in blended cement paste up to 15% of T-CCW, which may be attributed to
441 the limited pozzolanic activity of T-CCW [68] and the pores of T-CCW provided a favourable
442 environment for the growth of hydration products [57]. The XRD patterns show less intense

443 peaks at 32.2° and 41° for C_3S and C_2S , suggesting enhanced hydration kinetics. Furthermore,
444 the CC peaks are observed at 23° , 29.4° , 39.4° , and 43.2° . The water absorbed by the pores of
445 T-CCW facilitates the dissolution of CO_2 , leading to the consumption of CH generated during
446 hydration and subsequent carbonation to form CC [69]. The blended cement paste
447 demonstrated a higher degree of hydration compared to the control cement paste. This might
448 be due to the porous nature of T-CCW, which absorbs and retains moisture, and a micro filler
449 effect creating nucleation sites for the growth of hydration products within its pore structure
450 [57].

451 **4.4 Fourier Infrared Transform Spectroscopy**

452 Fig. 15(d) presents the FTIR spectra of the paste samples cured for 7 days. Compared with the
453 control sample (C0), no new absorption peaks were observed for the functional groups in the
454 T-CCW blended samples. These observations are consistent with XRD results. The absorption
455 peaks were observed at 3640 , 3400 , 1650 , 1416 , 1110 , 950 , and 870 cm^{-1} [66]. The absorption
456 peak at 3640 cm^{-1} is attributed to the stretching vibration of OH in CH, and the broadband
457 around 3400 cm^{-1} is associated with the stretching vibration of H_2O in hydration products,
458 representing chemically bound water molecules in the hydration product [70]. The intensity of
459 these bands decreases with increasing T-CCW, suggesting the presence of OH-phases. This
460 could indicate the consumption of pozzolanic reactions. These findings correlate with the TGA-
461 DTG curve calculated for the percentage of CH. The peak at 1650 cm^{-1} represents the hydroxyl
462 group (H_2O) bending, indicating consistent amounts of bound water in all the mixes. At 1416
463 cm^{-1} , stretching vibrations of CO_3^{2-} suggest the presence of carbonate phases. The increase in
464 peak intensity indicates a significant amount of CC in the carbonation products [71]. A minor
465 peak at 1110 cm^{-1} corresponds to the sulphates (SO_4^{2-}), indicating the presence of AFt or AFm
466 phases [72], which can be observed in SEM. The absorption peak at 950 cm^{-1} corresponds to
467 Si-O stretching, indicating the hydration product C-S-H [73]. The intense band at 870 cm^{-1} is
468 associated with the out-of-plane bending of CO_3^{2-} in carbonates [74], further supporting the
469 increased carbonation in the samples.



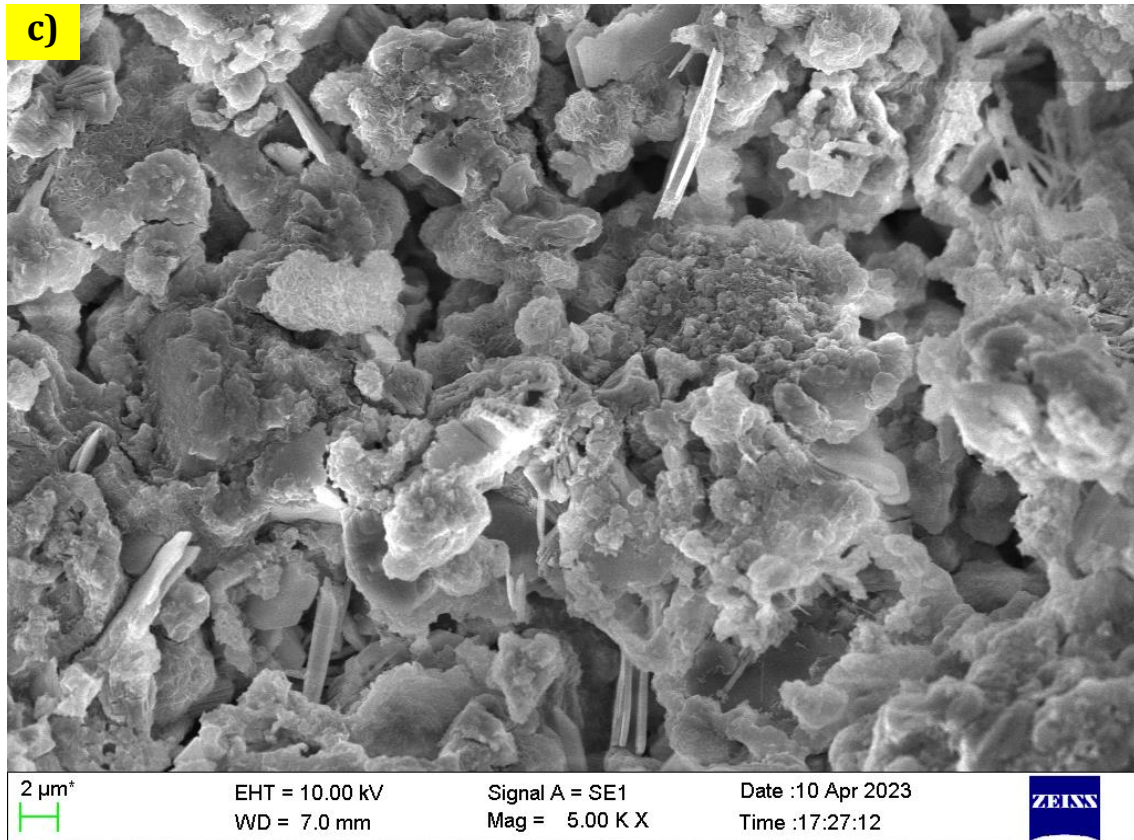
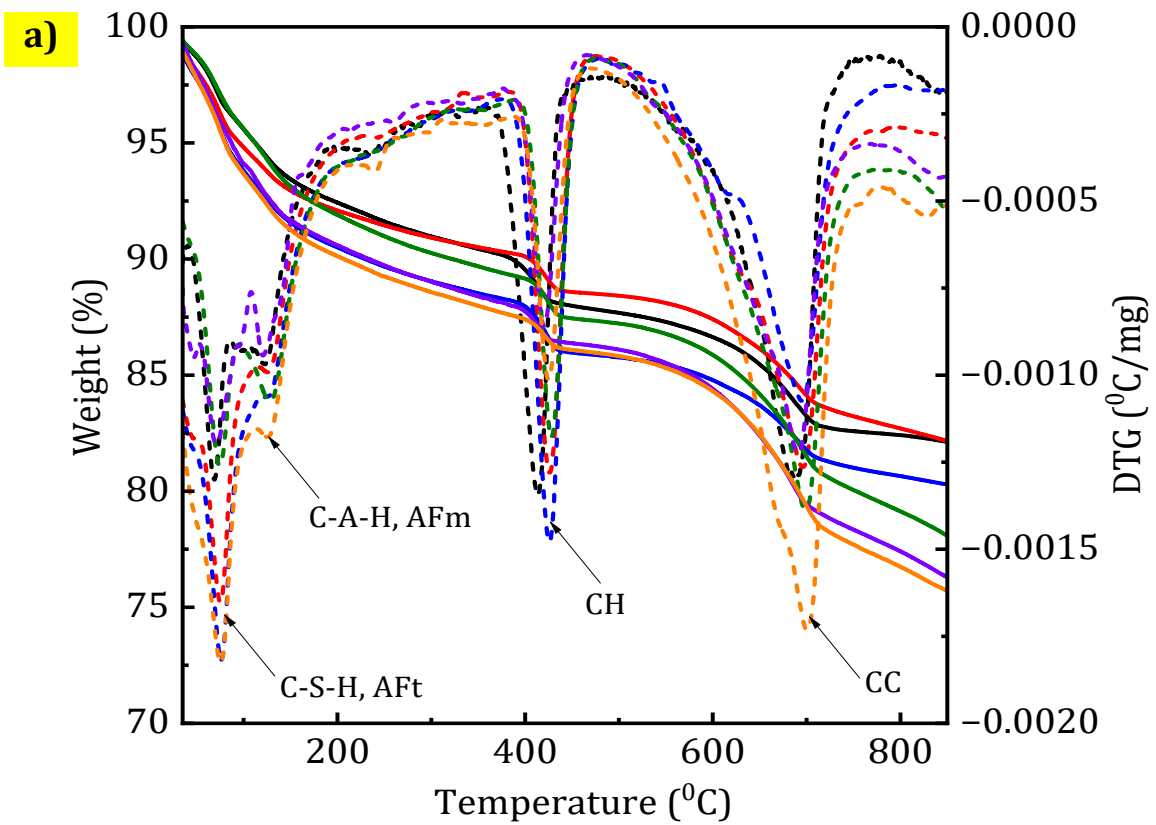
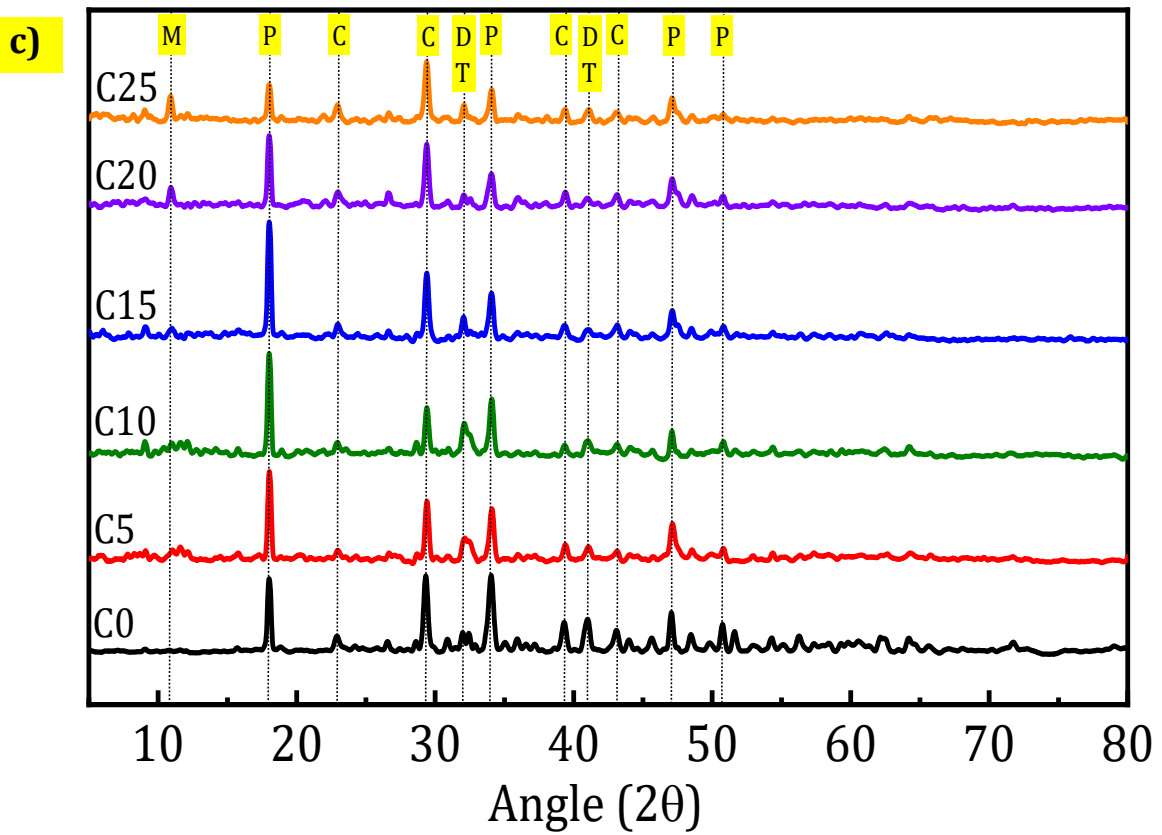
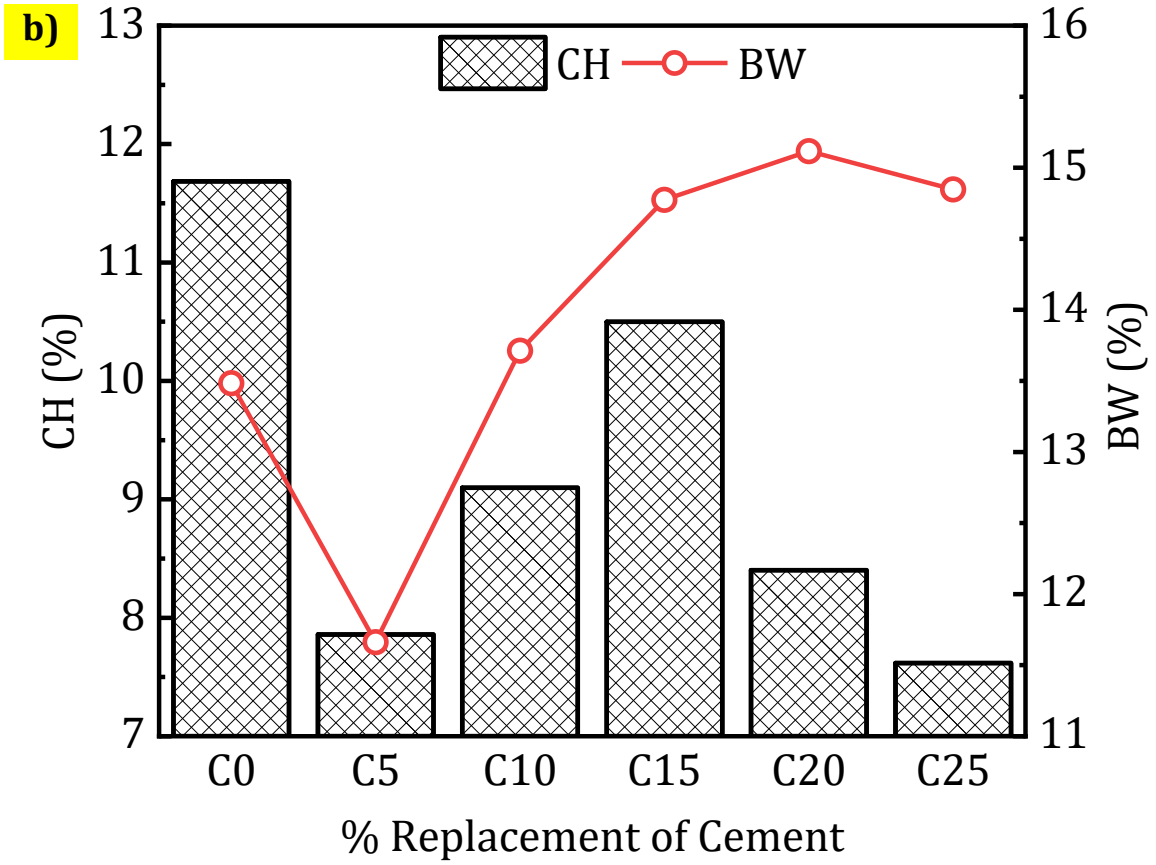


Fig. 14. SEM of the blended mixes at the T-CCW replacement of a) 0%, b) 10%, c) 25%

470





d)

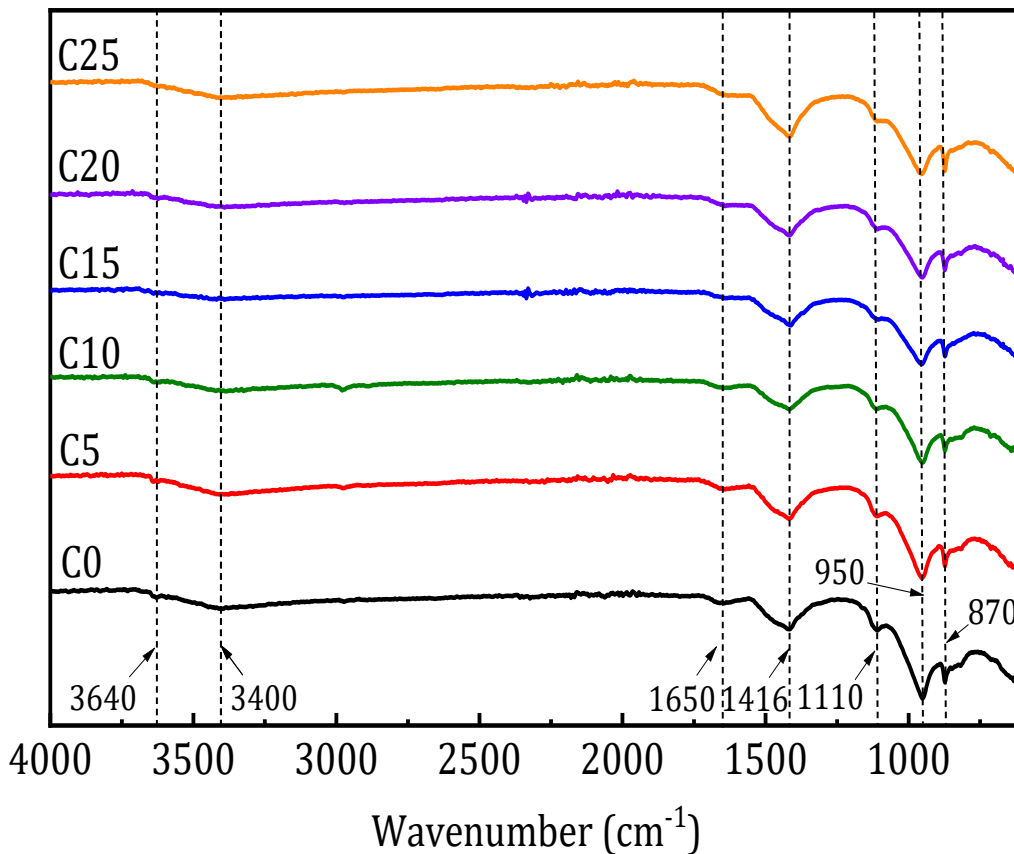


Fig. 15. a) TGA b) Percentage of BW and CH c) XRD d) FTIR of the blended pastes

471 5. Environmental Assessment

472 5.1 Sustainability Assessment

473 Researchers are exploring various alternative materials derived from industrial by-products and
474 agricultural wastes in the search for sustainability in the construction industry. These materials
475 not only substitute for conventional materials but also significantly lower the overall carbon
476 footprint of the production. This approach addresses the pressing need for sustainable
477 construction practices and tackles waste management challenges.

478 A simplified methodology is adopted to quantify the embodied carbon of the mortar
479 mixes based on the summation of the embodied carbon of each material. The embodied carbon
480 of the material, except T-CCW, was sourced from literature. The embodied carbon for cement
481 is 0.931 [75], sand is 8.08×10^{-4} [75], and water is 1.12×10^{-4} [75]. Due to the unavailability
482 of the embodied carbon for T-CCW from the literature, the CO₂ emission factors were analyzed
483 by estimating the emission during T-CCW preparation, including transportation, drying,
484 grinding, sieving, and calcination processes. The CO₂ emission factor for T-CCW is presented

485 in Table 4. The embodied carbon assessment for the blended mortar was calculated for 1 kg of
 486 mortar using Eq. (3).

487
$$ECO_{2e} = \sum CO_{2i} \times W_i \quad (3)$$

488 Where CO_{2i} is the carbon factor, and W_i is the weight of each material used.

489 **Table 4** Calculation of CO_2 emission factor for T-CCW

Material	Energy requirements for 1000 kg of T-CCW			Emission factor (Kg CO_2 /kWh)	Transportation of 1000 kg		Total emission (kg CO_2 /kg)
	Consumption (kWh)				Distance (Km)	Emission factor (kg CO_2 /km)	
	Drying	Grinding and Sieving	Calcination				
T-CCW	25 [76]	174.6 [77]	12	0.79	150	0.148	0.189

490 Fig. 16 illustrates the embodied carbon of the blended mortar mixes. The control mix exhibits
 491 the highest embodied carbon of 0.44 $kgCO_2/kg$. The total embodied carbon decreased as the
 492 percentage of T-CCW increased. The embodied carbon of the blended mortar is 4%, 8%, 12%,
 493 16%, and 20% lower than the control mortar. These results demonstrate that the addition of T-
 494 CCW as a partial replacement to cement substantially enhances the sustainability of the mortar
 495 mixes.

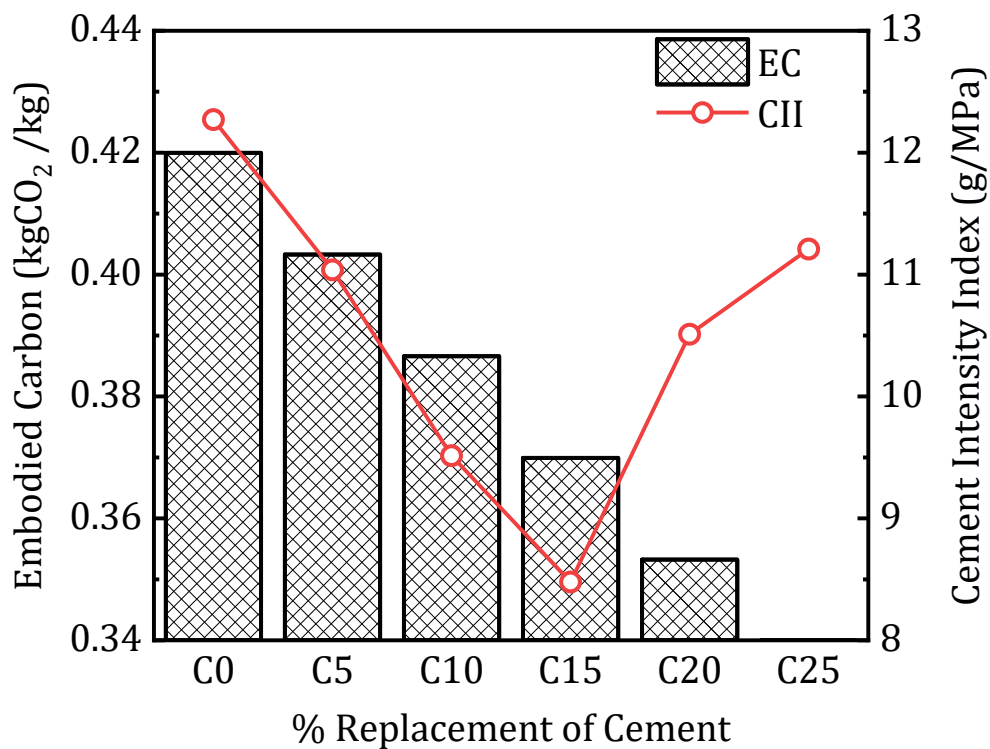


Fig. 16. Embodied carbon of blended mortars

496 In this study, CO₂ emission is the only environmental performance indicator used to
497 demonstrate eco-friendliness. To reduce the carbon footprint of T-CCW, the recycling approach
498 is to incorporate it into the mortar and mitigate the carbon emission of OPC. The cement
499 content accounts for 26.67% of the mortar ratio of 1:2.75. Therefore, the CO₂ emission of
500 cement in the control mortar is $(0.2667 \times 0.931) \approx 0.2483$ kgCO₂ per kg mortar. The CO₂
501 emission of T-CCW in the blended cement mortar, C25 is $((0.2667 \times 0.75 \times 0.931) + (0.2667$
502 $\times 0.25 \times 0.184)) \approx 0.1974$ kgCO₂ per kg mortar. Hence, the CO₂ reduction per kg of mortar is
503 $(0.2483 - 0.1974) \approx 0.0509$ kgCO₂ per kg mortar, approximately 20.5%. The amount of mortar
504 produced with 1 tonne of T-CCW, $1 / (0.2667 \times 0.25) \approx 15$ tonnes. The total CO₂ reduction per
505 tonne of T-CCW is $(0.0509 \times 15000) \approx 763.5$ kgCO₂ ≈ 0.7635 tonnes CO₂. This approach
506 primarily focuses on the direct carbon emission associated with material production. Further, a
507 comprehensive environmental assessment is required from cradle to gate to provide a more
508 holistic view of the material's environmental impact.

509 **5.2 Cement Intensity Index**

510 Fig. 16 presents the analysis of CII, indicating that the utilization of T-CCW as an alternative
511 to cement in mortar necessitates less cement to attain a strength of 1 MPa compared to the
512 control mix at 28 days. The control mortar exhibits a CII of 12.27 g/MPa, while the blended
513 cement mortar shows lower values ranging from 8.48 to 11.21 g/MPa, with the C15 mix being
514 the least compared to the control mortar, resulting in a saving of approximately 31% compared
515 to control mortar. The results indicate that it is possible to produce quality mortar with
516 comparable or better mechanical properties while significantly reducing cement content
517 relative to the control mortar [78].

518 **6. Factors affecting the utilization of T-CCW as a sustainable construction** 519 **material**

520 The fishbone diagram is a tool used to identify and explore the potential causes of a specific
521 problem. It also provides ideas for future work to enhance further understanding of the
522 problem. Fig. 17 presents a fishbone diagram illustrating the factors influencing the
523 performance of CCWA as a cementitious composite. This visual representation summarizes the
524 various parameters contributing to the effectiveness of CCWA in construction applications. The
525 diagram outlines eight primary factors: raw material, mix design, curing, testing, environment,
526 economy, and regulatory. These factors branch into sub-factors, highlighting the nature of
527 developing and implementing CCWA-based cementitious materials. This overview serves as a

528 roadmap for researchers and industry professionals, identifying key areas for investigation,
529 optimization, and potential challenges in the development and application of CCWA-based
530 cementitious composites. By addressing these factors, the potential of CCWA as a sustainable
531 alternative as a construction material can be fully understood.

532 **7. Conclusion**

533 Reducing, reusing, and recycling are crucial to achieving sustainability. Exploring alternative
534 materials from agricultural waste, such as CCW, generated from coffee processing addresses
535 waste management challenges while decreasing the use of energy-intensive materials. This
536 study examines the potential of T-CCW as a supplementary cementitious material, contributing
537 to the development of sustainable construction practices. The following conclusions can be
538 drawn from the study,

- 539 • The addition of T-CCW increased water demand for consistency, with water requirements
540 increasing up to 46% compared to the control mix at 25% replacement.
- 541 • The high surface area and porous structure of T-CCW material provide nucleation sites for,
542 and the water retention capacity of the T-CCW accelerated the setting times, with 25%
543 replacement of T-CCW reducing initial and final setting times by 73% and 63%,
544 respectively, compared to the control.
- 545 • The micro-filler effect and limited pozzolanic potential of T-CCW up to 15% improved the
546 compressive strength of mortar at all curing ages compared to the control mix. The 15% T-
547 CCW mix showed strength improvements of 82%, 53%, 49%, and 23% at 1, 3, 7, and 28
548 days respectively. The average compressive strength of T-CCW blended mortar was
549 significantly higher than the current minimum strength required for the masonry units.
- 550 • UPV test results indicated improved homogeneity and compactness in all the blended
551 cement mortars.
- 552 • Microstructural analysis showed that T-CCW enhances cement hydration and leads to
553 denser microstructure through pozzolanic reaction and improved nucleation.
- 554 • Environmental analysis revealed that incorporating T-CCW reduced the embodied carbon
555 of mortar mixes by 19% at 25% replacement.
- 556 • The cement intensity index was lowest for 15% T-CCW mix, indicating a 31% reduction in
557 cement needed to achieve 1 MPa strength compared to the control.
- 558 • T-CCW can be used as a partial replacement for cement, with optimal performance at 15%
559 replacement, contributing to cleaner production.

560 • Further research is needed to address various factors identified in the fishbone diagram to
561 optimize the use of CCW in cementitious composites.

562 • Thus, this research shows the use of T-CCW as a sustainable material in cementitious
563 composites when used in appropriate proportions.

564 Overall, incorporating T-CCW into mortar offers benefits across technical, economic, and
565 environmental perspectives. Technically, CCW can potentially improve strength and
566 microstructure properties. Economically, the use of T-CCW can lower material costs, reduce
567 waste management expenses, and create new opportunities in the green construction market.
568 Environmentally, utilization of T-CCW diverts waste from landfills, lowers the carbon footprint
569 associated with mortar production, and conserves natural resources. Thus, this research shows
570 the use of T-CCW as a sustainable material in cementitious composites when used in
571 appropriate proportions.

572 The study was limited to a specific calcination temperature; hence, future research could
573 investigate the effects of different calcination temperatures on the properties of CCW and its
574 effectiveness in cementitious composites. Research is underway to determine the influence of
575 T-CCW on the long-term durability properties, such as chemical attacks, carbonation, chloride
576 ingress, and corrosion, and to extend the possibility of using T-CCW in structural concrete.
577 Further, experimental work can be used to assess the leaching behaviour of T-CCW.
578 Additionally, the research could explore the application of T-CCW in various types of concrete
579 and its synergetic potential with other SCMs. Pilot scale trials and demonstrations are needed
580 to validate the performance of T-CCW in real-world applications. Additionally, evaluating the
581 economic feasibility and conducting a life cycle assessment of T-CCW production of large-
582 scale production is essential for encouraging its widespread use. Addressing these limitations
583 and pursuing these future research directions, significant progress can be made in optimizing
584 the utilization of T-CCW and validating its suitability as an SCM in cementitious composites,
585 thereby contributing to the circular economy strategies of the coffee processing industry.

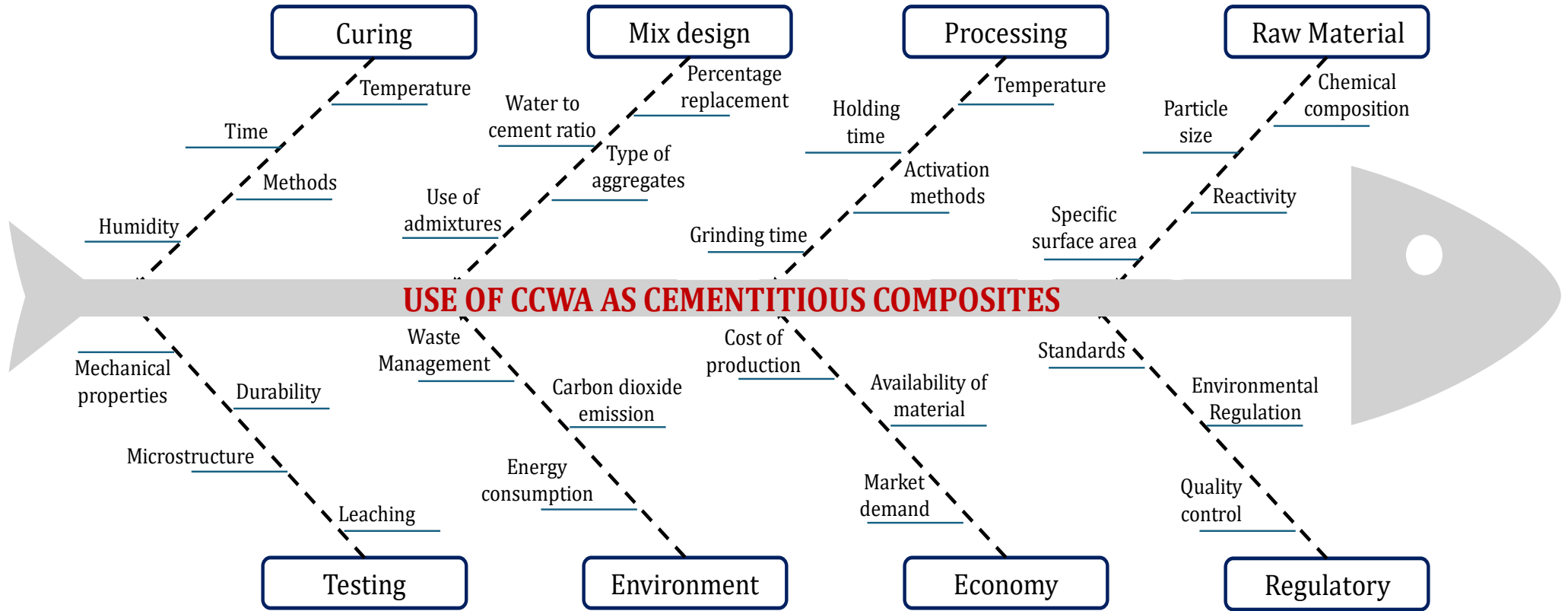


Fig. 17. Fishbone diagram

588 **Credit authorship contribution statement**

589 **Balasubramanya Manjunath:** Conceptualization, Investigation, Data curation, Validation,
590 Formal analysis, Methodology, Visualization, Writing - original draft, Writing - review &
591 editing. **Claudiane M. Ouellet-Plamondon:** Conceptualization, Data curation, Validation,
592 Writing - review & editing. **Anjali Ganesh:** Writing - review & editing, Funding acquisition.
593 **B.B. Das:** Data curation, Validation, Writing - review & editing. **Chandrasekhar Bhojaraju:**
594 Conceptualization, Investigation, Data curation, Validation, Formal analysis, Methodology,
595 Visualization, Supervision, Writing - original draft, Writing – review & editing, Funding
596 acquisition.

597 **Declaration of competing interest**

598 The authors declare that they have no known competing financial interests or personal
599 relationships that could have appeared to influence the work reported in this paper.

600 **Data availability**

601 No data was used for the research described in the article.

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