

19 mixtures containing GGBS. The results from the nomogram show a promising trend for the service
20 life of mixtures containing G, GO, and GGBS.

21 **Keywords:** graphene; graphene oxide; GGBS; rheology; ICP-OES; resistivity

22 **1. Introduction**

23 The addition of nanomaterials to cementitious materials presents an opportunity to achieve highly
24 durable and long-lasting materials. This can significantly increase the structural performance of
25 building materials [1]. Recent research has focused on the reliability-based durability design of the
26 nano-based cementitious material used in the construction industry [2-5]. To achieve this, building
27 design codes need to account for the properties of the proposed building materials. This requires
28 conducting different tests to provide details for addressing specific issues, i.e., rheological
29 properties, mechanical characteristics, durability, corrosion, and service life. These tests typically
30 require long time frames of decades to be conclusive [6].

31 Different nanomaterials have been used in cementitious composites, such as nano-silica [7, 8],
32 nano-alumina [9, 10], polycarboxylates [11, 12], nano-titanium oxide [13], nano-kaolin [14, 15],
33 nano clay, carbon nanotubes (CNT) [14, 16], and graphene-based nanomaterials (reduced
34 graphene oxide [17], graphene oxide [2] and graphene nanoplatelets [2]). Graphene-based
35 nanomaterials are the most innovative nanomaterials used to improve the characteristics of
36 concretes [17]. Graphene is a material of enormous scientific interest because it is a stable two-
37 dimensional material produced through cost-efficient chemical exfoliation techniques. Its planar
38 hexagonal lattice exhibits nuanced electrical properties, and its high aspect ratio lends itself to
39 many nanomechanical phenomena when used in composites [18]. In the past decade, further

40 advancements in the field have expanded the potential of graphene by developing similar materials
41 with different surface properties.

42 Graphene oxide and reduced graphene oxide contain partial surface coverage of hydroxyl and
43 epoxide functional groups [19, 20]. As a two-dimensional material, graphene has an exceedingly
44 high specific surface area. These minor chemical modifications dramatically impact the material
45 properties and their dispersion and bonding with other materials [21]. This is of particular
46 importance in applications in composites, such as in concretes, where the relative strength or
47 weakness of the interface can be leveraged to enhance the composite properties [22]. Graphene-
48 based nanomaterials have been shown to increase mechanical characteristics and durability in
49 cement [23]. Graphene primarily acts as a filler and improves the microstructure of cement paste
50 [24]. Graphene oxide arrests the formation of cracks, inhibiting crack propagation [25], and serves
51 as a nucleation agent for calcium silicate hydrate CSH [26]. Pan et al. [27] reported that using 0.05
52 wt% GO can increase the composite compressive strength by 15–33%. Peng et al. (2019) [28]
53 found that the addition of GO to cement paste improves the microstructure of cement hydration
54 products, refines the crystal size, and forms a denser and more uniform network structure.
55 Similarly, Lv et al. (2013) [29] confirmed that GO nanosheets could regulate the formation of
56 flower-like crystals and considerably improve the tensile/flexural strength. Preparation and
57 dispersion of these nanomaterials within the cementitious matrix have a critical effect on the
58 mechanical properties [30]. However, these nanoparticles reduce the workability characteristics of
59 cementitious pastes [24, 31-33]. Zohhadi (2015) [34] stated that in contrast to the improved
60 compressive strength reported by other authors at low dosages, increasing the concentration
61 graphene nanoparticles has a negative impact on workability which leads to a reduction in the
62 compressive strength. Li (2017) [31] observed the reduction of 21% in mini-slump diameter with

63 the addition of 0.03% of GO. Agglomeration of GO can lead to a large amount of entrapped water
64 in the cement.

65 To address these issues, few researchers used supplementary cementitious materials (SCMs),
66 including fly ash, silica fume, to compensate for the adverse effect of using nanoparticles on fresh
67 properties [35, 36]. Wang (2017) [36] tested fly ash for improving the rheological properties of
68 GO-modified cement-based materials. A decrease in the paste yield stress and plastic viscosity
69 was observed with the increasing fly ash content. At 0.01 wt% of GO and 20 wt% of fly ash, the
70 yield stress of the paste decreased 85.81%, and the plastic viscosity decreased 29.53% compared
71 to the control sample (without fly ash and GO). At 0.03 wt% of GO and 20 wt% of fly ash, the
72 yield stress of the paste is 50.3% lower, and the plastic viscosity decreased slightly by 5.6%. This
73 improvement in workability characteristics is mainly because of the ball effect, particle size
74 gradation, and lower water demand of fly ash [36]. Shang et al. (2015) [37] observed that the
75 addition of GO into the cement causes a significant reduction in fluidity and increases the
76 rheological parameters. They used silica fume as SCMs to compensate for the lower fluidity of
77 GO-based cementitious materials. Li et al. (2016) [38] reported that silica fume is also effective in
78 improving GO dispersion in cement paste. These prior studies in utilizing SCM to counteract the
79 fluidity reduction caused by the addition of nanoparticles to cement pastes demonstrates the
80 capacity for this methodology to be generalized to other SCMs.

81 Ground granulated blast furnace slag (GGBS) has been frequently studied as an SCM but has never
82 previously been considered as a potential additive to nano-modified cement pastes. The present
83 study seeks to investigate this potential for the purposes of counteracting the reduction in fluidity
84 caused by the addition of G and GO. In conventional concrete mixes, the addition of GGBS,
85 commonly in the range of 30-50%, has been shown to enhance the fluidity proportionally with the

86 GGBS content [39]. Regarding the affirmative influence of GGBS on concrete workability, a
87 similar promising trend was reported by the literature [40-45]. These rheological observations
88 correlate increased workability in the mortar with the increase in the surface area of GGBS and
89 the fluidity of mortar was linked to the morphology of GGBS. Smoother particle surfaces yielded
90 greater was the fluidity of the resulting mortar [46]. However, more experimental studies are
91 necessary to confirm these experimental observations with different SCMs, especially for
92 nanoconcrete mixtures. Hence, the current study applies GGBS to improve the fresh properties of
93 G- and GO-modified cementitious materials. Complementary research is performed to investigate
94 the effect of GGBS on service life characteristics of G- and GO-modified materials. The present
95 study intends to fill the current research gap and address the following questions.

- 96 1) To what extent do the G and GO influence the fresh properties of cement paste?
- 97 2) How effective is GGBS in improving the fresh properties of cement paste containing
98 nanomaterials?
- 99 3) What is the effect of nanomaterials on the service life of cementitious materials with
100 GGBS?

101 To address these questions, an extensive experimental program has been carried out. The present
102 study investigates different fresh properties, including (a) mini-slump spread diameter, (b)
103 viscosity, and (c) thixotropy. Moreover, the present study considers various mechanical and
104 durability tests to address the third objective, including the compressive strength, the conductivity
105 of the pore solution, and the concrete resistivity. As no prior research has focused on the pore
106 solution of mixtures containing G and GO, the present study fills this research gap. Moreover, the
107 service life of cementitious materials is dependent on the composition of the pore solution [47-50].

108 A complementary durability study is presented as the third objective to quantify the impact of the
109 pore solution changes on the durability of cement with G and GO.

110

111 **2. Experimental test**

112 ***2.1. Materials***

113 ASTM Type I Portland cement (PC) was used for all mixtures with a constant water-to-
114 cementitious material ratio of 0.35. The cement composition from quantitative X-ray diffraction is,
115 in percent weight, 56.68% C3S, 12.22% C2S, 6.31% C3A, and 11.00% C4AF. GGBS with the
116 chemical composition of 34.8% SiO₂, 8.5% Al₂O₃, 1.1% Fe₂O₃, 40.1% CaO, 9.7% MgO, 2.2%
117 SO₃, and 0.7% Na₂O were used at typical replacement levels (15%, 30%, and 45%). Volume-based
118 characteristic particle diameters of cement and GGBS were measured in isopropanol suspensions
119 by laser diffraction, and the values are reported in Table 1. A commercial polycarboxylate ether-
120 based polymer was used as the superplasticizer (SP) in the present investigation and was measured
121 by weight as a percentage of the cementitious materials. In ultrapure water, the graphene particles
122 were dispersed with a high-speed shear mixing mixer for one hour at 5000 rpm using sodium
123 cholate [51], and graphene oxide were dispersed (4 g/l) in commercially available ultra-pure water.
124 The dispersion quality is obtained by comparing TEM images along with visual observation.
125 Solution cholate is a surfactant used to better disperse graphene particles in water, while graphene
126 oxide performs better even without surfactant. Similarly, previous studies used sodium cholate as
127 a surfactant to disperse nanomaterials in cementitious materials [34, 51]. Although there is no
128 specific information regarding the influence of sodium cholate on the hydration process, it can
129 deduce that the concentration of sodium cholate within cement paste is approximately 0.003

130 percent showing a minor influence on hydration. Fig. 1(a) presents the transmission electron
131 microscopy (TEM) image of water-dispersed graphite with well-crystallized graphene sheets. Fig.
132 1(b) shows the topography of the commercially available GO. The thickness of the exfoliated
133 graphite flakes ranged from 150 to 200 nm which is in agreement with previous studies [52], while
134 the graphene oxide thickness is approximately 1 nm. The scanning electron microscopy (SEM)
135 image of GGBS is also shown in Fig. 1(c). The elemental compositions and the sizes of the
136 graphene and graphene oxide are presented in Table 2. These can significantly affect the properties
137 of the cement matrix. Fourteen different mix proportions were used in the present investigation to
138 form the basis of the new test results. Compositional parameters investigated were the dosages of
139 the nanomaterials and the supplementary cementitious materials. The mixture proportions are
140 given in Table 3. Local glacial sand with a density of 2,700 kg/m³ and a fineness modulus of 2.56
141 was used as fine aggregate for testing the hardened properties of cementitious materials.

142

143 ***2.2. Mixing procedure and specimen preparation***

144 The cement paste was prepared by mixing cement, GGBS (if applicable), water, superplasticizer,
145 and an aqueous dispersion of either graphene or graphene oxide with a constant water/cementitious
146 weight ratio of 0.35. The cement with a w/c ratio of 0.35 and 1% SP serves as the reference sample.
147 To investigate the influence of the nanomaterials, the dosage of G and GO nanosheets was varied
148 from 0.00% to 0.06% by weight of cement. The replacement ratio of cement with GGBS was
149 varied from 15% to 45% by weight (15%, 30%, and 45%). Table 3 presents the mix proportions
150 of cement paste and mortars used in the present investigation. To attain good dispersion of the
151 nanoparticles in the paste, a mixing procedure of ASTM C1738 [53] was adopted. Water and SP
152 were added to a high shear mixer with a water-cooling system that was selected to meet

153 specifications. The nanomaterial dispersions were added and stirred for 15 seconds. Then, the
154 cement was added within 60 seconds and the mixture was stirred at a speed of 4000 rpm for 60
155 seconds and 10000 rpm for 30 seconds. A 150-second rest was given, during which any paste
156 adhered to the sides of the bowl was reincorporated into the batch. Then, there was a final mixing
157 step of 10000 rpm for 30 seconds. Immediately after mixing, each of the mixtures was divided into
158 three plastic tubes; two for the mini-slump test and one for the rheometer.

159 For pore solution to ensure homogeneity in the raw materials, a blend was made for each of the
160 cementitious materials from samples taken from the storage sealed bucket at different heights. In
161 the first case, different nano dosages (0.03 and 0.06) were studied. In all the cases, the mixing was
162 performed uniformly (i.e. 2000 rpm for 60 s, rest for 30 s, and mix again at 2000 rpm for 90 s).
163 After mixing, the sample was rested at ambient temperature (25°C) for 15, 45, and 90 minutes. To
164 extract the pore solution, samples were placed in a centrifuge for 10 min at 5000 rpm. This
165 procedure was similarly used by the literature to obtain pore solution from fresh paste [54-56]. The
166 resulting supernatant solution was collected in a disposable plastic syringe and filtered through a
167 syringe filter of 0.45 µm. After filtration, the pore solution was immediately diluted 1:10 and 1:100
168 by mass with HNO₃ (w/w) in ultra-pure water for quantifying low- (Al, Fe, Mg, Si) and high-
169 concentration elements (Ca, K, Na, S), respectively. Diluted solutions were kept in screwcap
170 polypropylene containers, sealed with parafilm, and stored at 4°C until the test was performed.
171 The ICP analyses were carried out within one week of the extraction. The remaining undiluted
172 pore solution was used for measurements of the pH and conductivity.

173 Four concrete samples of each composition were prepared for resistivity and three samples for
174 compressive strength by mixing cement, GGBS (if applicable), fine aggregate, water, SP, and
175 nanomaterials (G and GO) with the surfactant. Initially, half of the water and the remaining SP

176 were mixed with water-dispersed nanomaterials (sonicated with half SP for 30 minutes before
177 mixing) until a homogenous solution was obtained. Then, the cement and the aggregate were added
178 to the mixture. Finally, the remaining water was added.

179

180 ***2.3. Test setups and procedures***

181 ***2.3.1 Mini-conical test***

182 For mini-slump test, a smaller version of the Abrams test was considered in the present study that
183 followed the same characteristic of the cone described in ASTM C1437 [57]. Freshly mixed
184 cement paste was filled in a truncated mini-cone (Top diameter 19 mm; bottom diameter 38 mm,
185 and height 57 mm) [58], which was placed on a grid marked plate (20 x 20 mm²). After removing
186 the filled cone Fig. 2, the cement paste flows and reaches a steady-stable state, and the increase in
187 the cone diameter is measured from a photo taken from the top of the set-up. This procedure was
188 repeated just before and after the rheometer test, as shown in Fig. 3. The diameter of the fresh paste
189 sample is the mean value of four measurements made in perpendicular directions.

190

191 ***2.3.2. Pre-shearing and flow test***

192 Flow curve testing was performed with an Anton Paar MCR 301 rotational rheometer. The
193 equipment was designed following the national institute of standards and technology (NIST)
194 recommendations [59]. The rheometer was configured with a double spiral spindle centered on a
195 metal shaft (Fig. 2). This tool has a diameter of 25 mm and a length of 55 mm from the bottom to
196 the top of the spiral. During the measurements, the temperature of the outer cylinder was
197 maintained at 23 ± 1.0 °C with a Peltier apparatus. Each cement paste sample was preconditioned

198 with a pre-shear of 0.1 rpm until an equilibrium state was achieved, followed by a 15 second rest
199 period. As shown in Fig. 3, the flow curve test was performed by increasing the shear rate from
200 0.1 rpm to 100 rpm, stepwise in 15 increments. The decreasing shear rate measurements were
201 performed in the range of 100 rpm to 0.1 rpm in 20 increments. For each incremental measurement,
202 the shear was maintained until an equilibrium shear stress was measured and the torque was
203 recorded at a constant rotational speed. For the calibration of the spindle, the NIST recommended
204 calibrated paste (Standard Reference Material (SRM) 2492 [60]) was used. To reset the shear
205 history of the paste the up-curve is used. The data of the down curves (decreasing shear rate) were
206 fit to the SRM 2492 certified values using the Bingham approach, as implemented in the SRM
207 2493 NIST certification [60]. Subsequently, the calibration factors $K\tau$ and $K\mu$ were computed with
208 the NIST Data Calibration Tool. These parameters were used to calculate the shear rate, the shear
209 stress, the yield stress and the viscosity of the cement pastes using the procedure explained by
210 NIST in [60].

211

212 **2.3.3 ICP-OES**

213 A Scientific Dual View ICP-OES (Thermo Scientific 5110) was used for the elemental analyses
214 of the pore solution. ASTM C1875 [61] was followed in the present study for ICP-OES method.
215 In the current investigation, the values provided by the literature [54, 62] were used as a starting
216 composition for calibration. Further, the composition of the calibration solution was modified with
217 the results of the initial series to ensure coverage of the entire range of concentrations measured.
218 By this procedure, the values were obtained by interpolation. The operating parameters of the ICP-
219 OES used for the analysis are presented in Table S1.

220

221

222 **2.3.4 Hardened properties**

223 The resistivity of the concrete strongly depends on the microstructure of the cement paste,
224 moisture, salt content, and temperature. The microstructure is influenced by factors such as w/c
225 ratio, degree of hydration, and the type and amount of SCM. In the present investigation, four
226 identical cylindrical samples of diameter 50 mm and length 100 mm were used to measure
227 resistivity. This property was measured by the direct two probe technique (bulk electrical
228 resistivity). This test was conducted in the presents study based on the recommendation of ASTM
229 C1202 [63]. To ensure a uniform current density during the measurement, brass plate electrodes
230 of the same size and shape as the end surfaces of the specimen were used. The care was taken to
231 ensure good contact between the flat ends of the cylindrical samples and the brass electrodes by
232 wrapping the electrodes in a damp cloth to uniform charge transfer. The plate electrodes were
233 firmly clamped on the specimen by a ‘C’ clamp and electrically insulated from their environment
234 by neoprene rubber pads. Until immediately before each measurement, the cylinders were stored
235 in a humid room for curing. The surfaces were gently cleaned to remove any dust or loose material,
236 and the dimensions of each specimen were recorded. The resistivity was calculated as follows.

$$\rho = RA/L \quad (1)$$

237 where ρ = Resistivity in kohm-cm, R = Resistance measured in kohms, A = Area of the contact
238 surface in cm^2 , and L = length between two electrodes in cm. After each of the curing periods,
239 three identical sample cubes of dimension 50 mm were removed from the environmental chamber.
240 The surface moisture on the specimens was removed with a cloth prior to their being weighed. The
241 cubes were subjected to unconfined compressive strength tests. A testing machine capable of
242 loading up to 2000 kN was used to apply the load at a compression rate of 1 kN/sec.

243 **3. Results and discussion**

244 **3.1 Workability**

245 **3.1.1 Mini slump**

246 To ensure the homogeneous properties in concrete, high flowability and moderate viscosity are
247 necessary for casting. These fresh properties assist in the proper compaction and shaping of
248 cement-based materials while maintaining adequate cohesion stability [27]. No signs of bleeding
249 were observed in any of these mixes. Fig. 5(a) and (c) show the variation mini-slump flow for the
250 cement pastes with different dosages of nanomaterials (G and GO) at each time. The mini-slump
251 diameter of the cement paste is approximately 220 mm at 7 minutes, and it was reduced to 205
252 mm at 22 minutes, yielding a slump of approximately 7%. Results show that as the percentage of
253 nanomaterials increase, the mini-slump value decreases. Identical trends were observed at 7 and
254 22 minutes for all the mixtures. The addition of 0.06% G and GO causes reductions of 50 and 75%
255 in the mini-slump values, respectively. Comparable results were observed at 22 minutes. This can
256 be attributed to the larger specific surface area of the G and GO. Similar results have been reported
257 elsewhere for cement [27, 32]. GO consists of oxygen-rich functional groups resulting in
258 agglomeration of cementitious materials and the formation of a flocculation structure which leads
259 to a further reduction in flowability [29, 64]. The effect of GGBS on the fluidity of the cement
260 paste with G and GO (at contents of 0.03 wt% and 0.06 wt%) is shown in Fig. 5(b) and (d). Results
261 show that adding GGBS, ranging from 15% to 45%, causes a considerable increase in mini-slump
262 diameters of nano-modified cement pastes. Similar promising results were observed for
263 unmodified cement paste containing more than 15% GGBS replacement (Fig. 5). At 15% GGBS,
264 the unmodified cement pastes were a little lower in mini-slump diameter than the ordinary cement

265 paste under the same solid volume fraction. This may be attributed to decrease packing density
266 due to the irregular shape of GGBS particles [65].
267 With the addition of GGBS, the mini-slump flow of the nano-modified cement improves. Increases
268 in flow diameter of approximately 31%, 109%, and 111% were observed for GO-mixtures
269 containing 15%, 30%, and 45% of GGBS, respectively. For graphene-modified cementitious
270 materials, 12%, 51%, and 135% increases were observed for the same dosages of GGBS. For G-
271 modified cement, dosages of 30% and 45% GGBS were found to yield flow diameters greater than
272 that of the control sample. However, for GO-modified cement, 45% of GGBS was required to
273 reach the control sample flow diameter. This demonstrates that the addition of GGBS can reduce
274 the flocculation in the cement pastes and increase the flow. Contrarily, due to the greater fineness
275 of the GGBS particles relative to cement (Table 1), the GGBS particles can enhance the particle
276 size gradation in the paste, which increases the fluidity of the nano-modified cementitious
277 materials [65].

278

279 ***3.1.2 Static yield stress***

280 Static yield stress indicates the degree of coagulation of concrete [66] which evolves with time as
281 thixotropy increases. If, in the flow curves, the rate at which the shear rate increases to its
282 maximum value is different than the rate which decreases from this value, a hysteresis loop will
283 form, as visualized in Fig. 6 [67]. The observed hysteresis is caused by the non-Newtonian paste
284 dynamics, which is dependent on the microstructure and composition of the material. As explained
285 graphically in Fig. 6, a shear thickening behavior corresponds to a thixotropic hysteresis area,
286 whereas shear thinning behavior results in a rheopectic hysteresis. In cementitious materials, the
287 static yield stress is measured as the peak shear strength at which interparticle attractive force

288 breakdown at low shear speeds. As shown in Fig. 7(a), the shear stress of nano-modified cement
289 pastes is generally higher than that of ordinary cement paste, which is due to the large surface area
290 of the nanomaterials. Moreover, the results indicate that the static yield stress of GO-modified
291 cement is higher than G equivalent at both dosages of 0.03 and 0.06 weight percentage of cement.
292 This may be due to the surface functionalization of GO which facilitates greater reactivity with
293 itself and other components of the cement paste, resulting in the agglomeration and the formation
294 of a flocculation structure [36]. It can be clearly seen in Fig. 7(a) that the cement pastes with higher
295 dosages of nanoparticles have greater yield stresses. This increase is more significant in GO-
296 modified cement paste as compared with G equivalents. As explained in more detail earlier, this
297 distinction between G and GO is primarily due to the surface functional groups in GO. Fig. 7(b)
298 illustrates the development of the static yield stress for different percentages of GGBS ranging
299 from 0% to 45%. The shear stress of GGBS-containing mixes is lower than that of the reference
300 mixture due to the reduced hydration between cement and GGBS. The low dosage of GGBS (15%)
301 has no significant influence on the yield stress (Fig. 7(b)). This can be attributed to the packing
302 density and irregular shape effects counteracting the reduced chemical activity of GGBS [65].
303 However, the impact of GGBS is pronounced at higher percentages (30% and 45%), because the
304 lower reactivity dominates the other phenomena [68].

305 The effect of the GGBS replacement in GO-modified cement paste is shown in Fig. 7(c). Results
306 demonstrate that 45% GGBS is the optimal dosage of GO-modified cement paste to mimic the
307 yield stress of the reference mixture. GGBS dosages of 15% to 30% showed substantial reductions
308 in yield stress compared to GO-modified cement alone but yielded higher yield stress than the
309 reference mixture. This promising finding can be attributed to the filler effect of GGBS in cement,
310 which counteracts the flocculation of GO with other components in GO-modified cement paste.

311 However, as there is less flocculation in G-modified cement paste, the GGBS replacement has a
312 greater impact on the fresh properties, and reference performance can be achieved at optimum
313 dosage of 30%. This can be mainly due to the reduction in porosity of G-contained cement paste.
314 In the case of 45% GGBS, a few signs of bleeding were observed in G-modified cement paste.
315 Fig. 8 summarizes the results of the hysteresis loop test. The first five columns give the results of
316 the nano-modified cement paste at different dosages. The latter nine columns show the impact of
317 GGBS-containing cement with and without nanomaterials. As expected, the addition of GGBS
318 and nanomaterials results in significantly lower hysteresis areas than the nanomaterials alone.
319 Results presented in Fig. 8 were calculated from the experimental data shown in Fig. 9. The
320 maximum thixotropic value of approximately 320 Nm/s was observed for the composition with
321 0.06 wt% GO due to the higher content of high surface area material, and GO's greater reactivity
322 due to its functionalization. In Fig. 9, the top two rows represent nanomaterial-modified cements,
323 while the lower rows are the mixes with GGBS and the nanomaterials. Each row is plotted on the
324 same scale to facilitate comparison. The blue shaded portion of the loop represents thixotropic
325 values, and red represents rheopetic which are directly compared in Fig. 8. All set of mixes showed
326 thixotropic behavior except for the reference, 0.03 G and 15 GGBS, which expressed both
327 thixotropic and rheopetic values (plotted as negative values). This represents that for these
328 mixtures, after the rigorous shearing, the downward shear plot had higher torque values than the
329 upward shear plot. A similar result was reported by Chhabra et al. [67] for materials like thick
330 suspensions of kaolin and cornflour in water. For such materials, the equilibrium concerning the
331 rate of structural rebuild and breakdown can tilt to either side depending primarily on the setting
332 conditions of the samples. However, a few researchers have cautioned that negative slopes in the
333 flow curve can occur when there is insufficient pre-shear to achieve a steady-state.

334 *3.1.3 Dynamic yield stress and viscosity*

335 In handling cementitious pastes, the dynamic yield stress is crucial to quantify the workability
336 [69]. The influence of the nanomaterials (G and GO) and GGBS modified nano-cementitious
337 materials on the rheological behavior of cement pastes, as studied in this work, can be seen in Fig.
338 10(b). There is an evident increase in the yield stress of the pastes at early testing times when
339 nanomaterials are incorporated into the mixture. Large agglomerates of precipitated graphene and
340 graphene oxide occupy a portion of the free water to wet the surface of nanomaterials. This results
341 in an adverse effect on workability, which has been acknowledged in the literature [64]. The further
342 enhancement of dynamic yield stress and viscosity caused by the addition of GO is observed from
343 Fig. 10 (a and c). The additional decrease of fluidity and increase of viscosity, compared to G-
344 modified equivalents, may be attributed to the greater dispersion and relative hydrophilicity of GO.
345 This results in a higher specific surface area for GO, increasing its interactions with cement
346 particles and promoting aggregation of cement grains. The functional groups on the surfaces of the
347 GO are reduced, transferring oxygen into the cement paste, accelerating the hydration of cement
348 and further reducing the workability [64]. Therefore, the use of other additives, such as fly ash,
349 GGBS, silica fume, and superplasticizers, has already been recommended in the literature to offset
350 the reduction in fluidity [37, 70]. Fig. 10 (b and d) shows the viscosity and dynamic yield stress
351 measurements for nanomaterial-modified cement paste with different dosages of GGBS (0%,15%,
352 30%, and 45%). The viscosity of the reference cement was approximately 0.207 Pa-s. When 15%,
353 30%, and 45% by weight GGBS is added, the value of viscosity was measured to be around 0.219
354 Pa-s, 0.202 Pa-s, 0.140 Pa-s, respectively. As described earlier, the 15% GGBS unmodified paste
355 experiences competing phenomena from the GGBS to modify the rheological properties, resulting
356 in a mild increase in viscosity. However, the higher dosages of GGBS yield the expected reduction

357 in viscosity. With the addition of the G and GO in the 15% GGBS paste, the viscosity further
358 increases. However, for G-modified cement pastes, the 30% and 45% GGBS yielded viscosities
359 lower than the reference. For GO, 45% of GGBS was required to attain viscosity values
360 comparable to that of ordinary cement paste. These results are consistent with observations of the
361 addition of other admixtures [70]. The same trends were also noted for the measurements of the
362 dynamic yield stress.

363 The data presented previously enables assessing the relationship between mini-slump values and
364 Bingham constants, including yield stress and plastic viscosity [71]. Fig. 11 shows the relation
365 between the slump-flow diameter and the mini-slump of the nano-modified cementitious materials
366 before and after the rheometry testing. The slump is correlated inversely with the yield stress by
367 an exponential relationship. The data fit shows a correlation coefficient of $R^2 = 0.96$. An inverse
368 linear relationship correlates the viscosity with an $R^2 = 0.80$. Many researchers have demonstrated
369 the relationships between mini-slump flow, viscosity, and yield stress [72-74]. As expected, the
370 addition of nanomaterials induces a rise in viscosity and yield stress (inversely correlated with
371 slump flow) due to a reduction in effective free water by the mechanisms described previously.
372 This trend increases with the quantity of nanomaterials as the number of contact points between
373 the particles leads to greater internal friction in the paste during flow, further increasing viscosity
374 and dynamic yield stress [72, 75, 76]. As shown in Fig. 11, the addition of nanomaterials increases
375 the plastic viscosity linearly along with a decrease in the mini-slump value [77]. This is attributed
376 to the high nanoparticle surface area, leading to greater Vander Waals bonding and flocculation of
377 G and GO. Bentz et al. [78] observed a similar linear trend for the plastic viscosity to the mini-
378 slump. The relationship between the dynamic yield stress and the mini-slump followed an
379 exponential correlation [79]. However, the deviation of these results between different periods

380 makes it evident that many factors influence the measured slump flow, including the hydration
381 kinetics and differences in the concentrations of Ca, Si, Na, K, which will be explained in the next
382 section.

383

384 ***3.2 Impact of nano-modified cementitious materials on pore solution***

385 The ICP-OES compositional analysis of the pore solution of each mixture at different times is
386 shown in Table 4. The results show a slight increase in calcium concentration in the pore solution
387 upon the addition of nanomaterials and GGBS, at low dosages of both. The ratio between the high-
388 concentration elements (such as K/Na) is unchanged by the addition of nanomaterials. However,
389 there is a drastic change in these proportions with the addition of GGBS, leading to a reduction in
390 the electrical conductivity of the pore solution of GGBS concretes. Table 4 also confirms a sharp
391 increase in low-concentration elements upon the addition of nanoparticles at low dosage, compared
392 to the reference cement. The same increase is visible for the nano-modified GGBS concretes. This
393 correlates well to the enhancement of compressive strength in these concretes.

394 Fig. 12 shows the concentrations in pore solutions of Ca, K, Na, S, Si, and Al of cement modified
395 with nanomaterials and GGBS at different dosages. Generally, the results show that the addition
396 of nanomaterials causes comparable and slightly elevated pore solution concentrations of these
397 elements compared to reference mixtures. Interestingly, the impact of GO appears to change at the
398 higher dosage tested; switching in some cases from causing a mild increase to a minor decrease in
399 concentrations measured. This can be attributed to the agglomeration of GO at the higher content
400 causing a reduction in the effective nanomaterial content which is consistent with the reduced
401 compressive strength and yield stress measured for these formulations. The addition of GGBS
402 caused decreases in the observed concentrations of Na, K, and S, while simultaneously increasing

403 the Si and Al contents and leaving the Ca unchanged. These observations are consistent with
404 expectations from the chemical composition of GGBS. However, the results for the nano-modified
405 mixtures with GGBS were varied by the element species. For Ca, Si, and Al the concentrations
406 were enhanced compared to the equivalent mixtures with GGBS alone. In contrast, lower pore
407 solution concentrations were observed for the other elements. This offers some explanation for the
408 observed increase in hardened compressive strength with a simultaneous reduction in paste yield
409 stress observed for these formulations.

410 The measurements of the electrical conductivity of the pore solution are slightly higher for nano-
411 modified materials. These results are shown in Table 4. The formation factor is directly
412 proportional to the conductivity of the pore solution [80]. For safety, the conductivity is assumed
413 to be constant for each group of nano-modified mixes. In the present investigation, the pore
414 solution resistivity is estimated based on binder chemistry, mix proportions [81, 82]. 80% degree
415 of hydration is assumed for all the mixtures at 28 days [80]. A value of 18.47 (s/m) is considered
416 for plain and nano-modified cementitious composites. Values of 16.05, 13.55, 10.95 (s/m) are
417 considered for 15, 30, 45% of GGBS modified cementitious composites, respectively.

418

419 ***3.3 Impact of nano-modified cementitious materials on the resistivity of concretes***

420 The durability of concrete strongly depends on the properties of its microstructure, such as pore
421 size distribution and the shape of the pore connections. Fig. 14(a) compares the results of the
422 resistivity of the nano-modified cement with the reference. With the addition of nanomaterials,
423 there is a slight decrease in resistivity. However, the resistivity of concretes increases drastically
424 with the addition of GGBS, as shown in Fig. 14(b). This effect was observed to increase with age
425 and dosage to a dramatic extent. Fig. 14 (c,d) presents the change in the resistivity of the GGBS

426 concretes with the addition of G and GO. From these results, it is clear that the dominant effect is
427 that of GGBS. Compared to the unmodified equivalent cements, the nano modified GGBS cements
428 show slightly lower resistivity at all ages except on the first day. This may be due to the enhanced
429 hydration rate cause by the addition of the nanomaterials. However, for all the resistivity at the 28
430 days, the corrosion probable rate can be applied as per the limitations suggested by Browne et al.
431 [\[83\]](#).

432

433 ***3.3.1 Calculation of formation factor***

434 The formation factor is the ratio of the bulk resistivity of concrete formulation to the resistivity of
435 the pore solution of a cement paste. It was proposed by Archie [\[84\]](#) to characterize the pore
436 structure of a porous rock and similar materials. The value strongly depends on the pore geometry
437 and connectivity of the pore network. Extensive work on concrete was conducted by Weiss and
438 Barrett [\[85, 86\]](#) on the formation factor of concrete structures. Fig. 15 compares the calculated
439 formation factors for nano-modified and GGBS-containing cement formulations considered in the
440 present investigation. With the addition of nanomaterials, there is a slight decrease in the formation
441 factor due to the conductive nature of the additives. Conversely, the formation factor increases
442 drastically with the addition of GGBS. However, for combinations of nanomaterials and GGBS,
443 all of the formation factors are significantly improved, and the corresponding corrosion probable
444 rates are very low as per W. Jason Weiss et al. [\[86\]](#).

445

446 ***3.4 Compressive strength of concrete***

447 Fig. 13 shows the 28-day compressive strengths of the cement formulations with and without
448 nanomaterials and GGBS. Without GGBS, the addition of 0.03 and 0.06 wt% of G and GO resulted

449 in significant increases in the compressive strength. This is in agreement with previous reports on
450 the effect of carbon-based nanomaterials on strength. However, in this work, it was observed that
451 the extent of strength improvement was reduced at the higher dosage of G and GO (0.06 wt%). As
452 discussed previously, this may be attributed to the agglomeration of the nanoparticles, reducing
453 their effective content in the composite. The addition of 15, 30, and 45 wt% GGBS also produced
454 substantial increases in compressive strength. Notably, the addition of nanomaterials only
455 increased the strength at the low dosage of GGBS. Higher contents showed significantly less
456 improvement. Furthermore, it was observed that G offers the greatest strength improvements with
457 low or no GGBS. Surprisingly, the compressive strength was found to increase with GO for the
458 highest content of GGBS, in contrast to the general trend.

459

460 ***3.5 Estimation of the service life of concrete***

461 To durability-based design a RC structure, the resistance of the concrete cover necessities to be
462 measured. This section intends to work on a monograph providing empirical equation for
463 evaluating this resistance in terms of the rate of chloride ingress through concrete, and therefore
464 the time to initiate the reinforcement corrosion. Once the formation factor for a cement
465 composition has been determined, the diffusion of chloride due to the permeability of the concrete
466 can be predicted by Flick's second law of diffusion [87], given in Eq. (2). This equation can be
467 solved for the parameter t , time, to yield a surface of solutions for possible inputs of the other
468 parameters. These solutions are depicted graphically in Fig. 16. In the case of predicting the time
469 of concrete failure, the C_x is the critical chloride concentration, C_s is the constant surface
470 concentration, C_o is the native chloride concentration within the cement which is assumed to be
471 zero, X is the depth of the reinforcement, D_o is the self-diffusion coefficient constant for a chloride

472 ion ($2.03 \times 10^{-9} \text{ m}^2/\text{s}$ at 25°C), F is the calculated formation factor, and t is the time. The time-
473 to-failure of concrete has been suggested by previous literature to be six years to be the parameter
474 t plus an initiation period of six years [86].

$$\frac{(C_x - C_0)}{(C_s - C_0)} = 1 - \operatorname{erf} \left[\frac{x}{2\sqrt{\left(\frac{D_0}{F}\right)t}} \right] \quad (2)$$

475 Fig. 16 provides a nomogram of the solutions for parameter t for typical combinations of formation
476 factors, chloride surface concentration, critical concentration, and the depth of reinforcement. To
477 interpret this graphic, extend a line perpendicular to the formation factor axis at the given formation
478 factor value through the surface concentration quadrant (Fig. S1 in Supplementary Materials). The
479 solution for t is then measured in years on the axis perpendicular to this line at the point of
480 intersection with the line which corresponds to the input surface concentration. This process can
481 be repeated from this intersection point into the depth of reinforcement. Subsequently, the critical
482 concentration quadrants to calculate the solution for t with alternate values for these parameters
483 given on the corresponding lines in each section. Default values are assumed in this calculation to
484 be 25 mm and 0.15, respectively.

485 It is worth mentioning that some limitations exist in the present study which needs to be considered
486 for future works, including microstructural analysis of GGBS-contained nanoconcrete. For
487 instance, the section of “estimation of service life of concrete” needs more experimental proof as
488 the definition of concrete failure by chloride ions is not completely clear. Since concrete
489 deterioration caused by NaCl, CaCl₂ and MgCl₂ have different mechanisms, it is necessary to
490 conduct the freeze-thaw durability test (ASTM C666) or salt scaling test (ASTM C672) to provide
491 some direct evidence showing the significant benefit of using GGBFS and G/GO.

492

493

494 **4. Conclusions**

495 An experimental program was carried out in the present study to determine the effect of graphene
496 and graphene oxide on fresh paste, hardened, and durability properties of cementitious materials
497 with varying contents of ground blast furnace slag. Mini slump, rheometry, compressive strength,
498 ICP-OES pore solution analysis, and electrical resistivity tests were conducted in the present study,
499 and the analysis of the results has yielded the following conclusions:

- 500 - Generally, results show that the addition of G and GO reduces the mini-slump flow
501 diameter of the cement paste. The content of the nanomaterials has a considerable effect
502 on this phenomenon, with 50% and 75% reductions observed for 0.06% of G and 0.06%
503 of GO, respectively. Due to the agglomeration, GO has a higher impact on the fluidity of
504 the mixtures as compared to G. Results show that adding GGBS to mixtures considerably
505 increases the flow diameter, especially for dosages of at least 30 wt%. Furthermore, GGBS
506 is more efficient in improving flow diameter in GO-modified mixtures as compared to G-
507 modified mixtures and reference.
- 508 - The hysteresis tests show that 0.06% GO has the greatest thixotropy compared to other
509 mixes, while a mixture containing 0.03% G is the most rheopitic. Results also show that
510 adding GGBS to the mixtures containing G induces greater thixotropic characteristics.
511 However, an adverse trend was observed for mixtures containing GO.
- 512 - Results show that adding G and GO results in higher viscosity and dynamic yield stress of
513 mixtures. This effect is more potent for GO-modified mixtures than G. Adding GGBS

514 alone has no considerable impact on the viscosity and the dynamic yield stress. However,
515 in the presence of G and GO, the addition of GGBS yields a significant reduction in these
516 properties. The results indicate that reference rheological behavior can be recreated for G-
517 modified mixtures by adding 30% of GGBS and GO mixtures with 45% GGBS.

518 - Results indicate that adding both G and GO creates higher compressive strength of mortars
519 compared to the reference mixture. However, a reduction in this effect was observed at
520 higher dosages. The addition of GGBS was shown to improve the compressive of mortars
521 containing nanomaterials. Generally, the trend indicates that mixtures containing G have
522 higher compressive strengths as compared to GO ones. The mixture with 0.03% G and
523 30% GGBS yielded the highest compressive strength among the mixtures tested.

524 - Results indicate that the addition of nanomaterials slightly decreases the resistivity of the
525 cements at 28 days. However, adding GGBS causes a considerable increase in 28-day
526 resistivities. Similar results were observed for G- and GO-modified mixtures. Additionally,
527 results indicate that mixtures containing 0.03% GO and 45% GGBS have the highest
528 formation factor among the compositions tested.

529

530

531 **CRedit authorship contribution statement**

532 **Chandrasekhar Bhojaraju:** Conceptualization, Idea, Investigation, Data curation, Validation,
533 Formal analysis, Methodology, Visualization, Writing - original draft, Writing - review & editing.

534 **Seyed Sina Mousavi:** Investigation, Formal analysis, Data curation, Writing - original draft,
535 Writing - review & editing. **Victor Brial:** Investigation. **Michael DiMare:** Writing - review &

536 editing. **Claudiane M. Ouellet-Plamondon:** Supervision, Validation, Resources, Funding
537 acquisition, Writing - review & editing.

538

539 **Declaration of competing interest**

540 The authors declare that there are no competing interests regarding the publication of this paper.

541

542 **Acknowledgements**

543 The authors would like to thank the ÉTS Montréal technical staff for support in our experiments.

544 The Natural Sciences and Engineering Research Council (NSERC) of Canada and Quebec Centre

545 for Advanced Materials (QCAM) supported this study. Moreover, the authors would like to thank

546 Ciment Québec Inc. and NanoXplore Inc. for supplying cement and graphene, respectively, for the

547 present study.

548

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