

# IMPACT OF HYDROGEN-DIESEL DUAL FUEL COMBUSTION ON THE PERFORMANCE OF AFTERTREATMENT SYSTEM IN HEAVY DUTY COMMERCIAL TRUCK

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**Abstract**— Hydrogen has emerged as a promising near-term solution for enhancing the performance of internal combustion engines. Hydrogen-diesel dual fuel (H<sub>2</sub>-DF), where hydrogen is introduced in the intake of a heavy-duty diesel engine, reduces tailpipe CO<sub>2</sub> emissions but also impacts other pollutants. This study investigates the correlation between the hydrogen addition to diesel engines and the effectiveness of exhaust aftertreatment systems (EATS) in reducing emissions. H<sub>2</sub>-DF combustion impacts the temperature and composition of the exhaust gas, which may impact the performance of EATS. To evaluate the performance of EATS components, analysis of emission data measured using a class-8 commercial truck operated on a chassis dynamometer at steady state conditions is conducted. Conventional diesel and H<sub>2</sub>-DF operation are compared at 1200 RPM and 200 Nm shaft torque. Engine-out emissions, NO<sub>x</sub> conversion rate, and temperatures at key points in the EATS are compared. The time for the EATS components to reach stable temperatures depended on the prior operating condition. In H<sub>2</sub>-DF, a significant increase in temperature across the catalyst system was observed compared to diesel, resulting in an increase in NO<sub>x</sub> conversion rates. In H<sub>2</sub>-DF, engine-out NO<sub>x</sub> emissions were comparable to conventional diesel while CO and CO<sub>2</sub> were reduced. A sensitivity analysis of diesel-based SCR with the computational tool GT-SUITE was also performed, which indicated SCR temperature and NH<sub>3</sub> dosing have significant impact on the performance of EATS.

**Keywords**—component; formatting; style; styling; insert (key words)

## I. INTRODUCTION

Diesel engines used in commercial trucking are one of the largest emissions sources from the transportation sector [1]. They emit pollutants including oxides of nitrogen (NO<sub>x</sub>) and particulate matter (PM) and greenhouse gases (GHGs) [2]. NO<sub>x</sub>

have harmful effects on the environment and human health [3], and are facing increasingly stringent regulatory requirements for emissions control. To cope with current and future regulations, modern diesel engines are equipped with an exhaust aftertreatment system (EATS) with multiple catalytic components to treat the different pollutants in the exhaust [4]. Adding hydrogen to the intake air in a diesel engine can reduce tailpipe CO<sub>2</sub> emissions, but its impact on the performance of the EATS needs to be investigated.

Typical heavy-duty diesel engine EATS consists of a Diesel oxidation catalyst (DOC), a Diesel Particulate Filter (DPF), and a Selective Catalytic Reducer (SCR), as shown in Fig.1. The DOC oxidizes unreacted HC and CO into CO<sub>2</sub>, while the DPF removes PM, including soot. The SCR converts NO<sub>x</sub> into N<sub>2</sub>: to achieve this reduction reaction in an oxygen-rich environment requires the addition of a reducing agent: Diesel Exhaust Fluid (DEF), a urea solution [5]. DEF is dosed at the upstream of SCR where it gets mixed with exhaust gas as depicted in the Fig.1.

The performance of EATS components is dependent on catalyst type and size and various operational factors, including exhaust gas composition, temperature variations, and space velocity, which can affect their efficiency. These factors influence the removal of pollutants due to chemical reactions taking place inside the individual EATS components.

A hydrogen-diesel dual fuel engine (H<sub>2</sub>-DF) engine involves injecting hydrogen upstream of the intake manifold, where it mixes with air before entering the engine cylinder, followed by the combustion of the air-gas mixture initiated by the diesel injection [6]. Hydrogen addition into the diesel engine may result in some unreacted hydrogen (hydrogen slip) in the exhaust gas as well as different compositions of exhaust gas due to its variations in the combustion process. The addition of hydrogen to diesel engines introduces unique dynamics that could impact the performance of EATS system. It changes the thermal conditions and the chemical reactions inside the system.

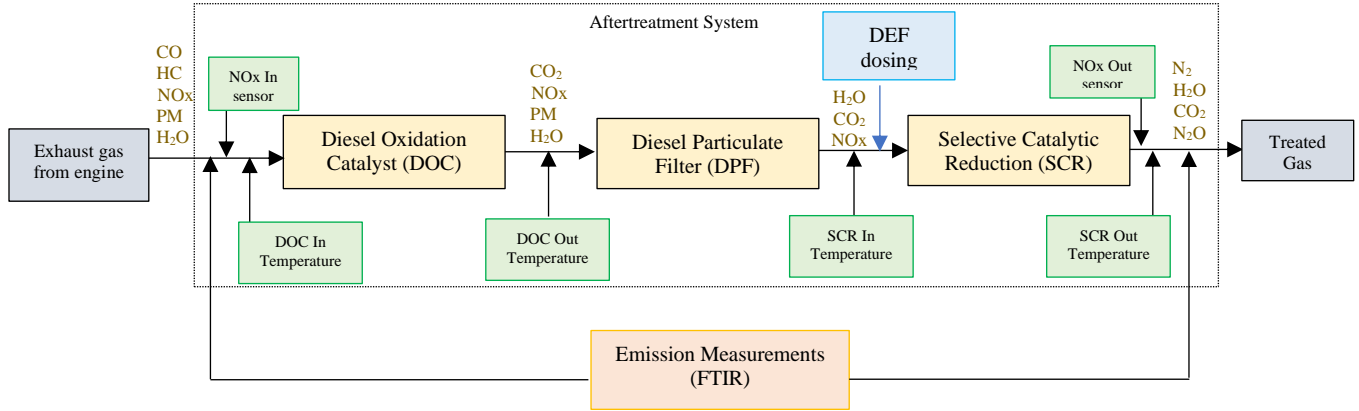


Figure 1: Block diagram of Aftertreatment system

The presence of  $H_2$  has been shown to enhance the performance of the DOC by promoting the oxidation of CO and HC [7].  $H_2$  also helps to oxidize NO into  $NO_2$  in the DOC, improving the performance of SCR as well. In the SCR, NO and  $NO_2$  react with  $NH_3$  from the DEF. The SCR reactions that involve both NO and  $NO_2$  is faster; thus, the rate of NOx reduction is higher in case of a higher  $NO_2/NO_x$  ratio. In SCR,  $H_2$  helps to reduce NOx and improve its performance [8] [9]. Any  $H_2$  present in the SCR would act as the reducing agent and react with NOx. However, it is likely that most  $H_2$  in the exhaust will have been oxidized in the DOC (see Fig. 1) and so would not reach the SCR.

The objective of this study is to understand the impact of  $H_2$ -DF on the EATS of a heavy-duty commercial truck, including the temperature of the primary EATS components and the NOx conversion rate. The analysis is based on chassis-dynamometer data combined with a simplified SCR model to assess sensitivity of the  $H_2$ -DF SCR at a low-load operating condition.

## II. METHODOLOGY

Hydra Energy has developed a  $H_2$ -diesel dual fuel dosing system for class-8 commercial heavy-duty trucks. The  $H_2$ -DF system performance was assessed by running a commercial heavy-duty truck on a chassis dynamometer in both diesel and  $H_2$ -DF modes under steady-state conditions [6]. Specifications for the vehicle, engine and aftertreatments system are provided in Table 1. Emissions were measured using a Fourier transform infrared (FTIR) system (Bruker MATRIX MG5), configured to measure from either upstream or downstream of the aftertreatment system. Onboard OEM NOx sensors, located

before the DOC and after the SCR, provided simultaneous measurements to supplement the FTIR. Temperatures through the aftertreatment components were also monitored using on-vehicle sensors. Reference [6] provides more information on the instrumentation and data acquisition procedures. The purpose of this study was to compare the emissions produced by the engine and the performance of EATS both for diesel and  $H_2$ -DF engine.

EATS temperatures were evaluated for investigation of their stabilization phase. The stabilization criteria for EATS components used in this study is characterized by minimal temperature fluctuations and a consistent temperature gradient across the component. For SCR, when its inlet temperature stabilizes and differential across the component is within  $10^\circ C$ , this condition is also considered as stable for this study. Stabilization time of these components is also dependent on the previous condition of engine and EATS. The prior condition of this test point for diesel fuel was  $H_2$ -DF operation at 14% engine load while for  $H_2$ -DF test point, engine was operating with  $H_2$ -DF with varying engine load under 25%.

The  $H_2$ -DF condition was achieved by adding  $H_2$  and reducing the engine demand torque correspondingly to maintain a constant output shaft torque. The data point used for this paper was taken at engine speed of 1200 RPM and 200 Nm of shaft torque both for diesel and  $H_2$ -dual fuel. This point was chosen due to the higher complexity of the EATS performance at low engine load, due to low temperatures and flow rates. Along with experimental work, study through modelling is also carried out.

To supplement the experimental results, an SCR model for a diesel engine was developed in GT-SUITE, informed by literature data and emissions measurements. SCR modelling requires reaction kinetics as well as operating parameters. The most important component of SCR is the catalyst that promotes NOx reduction. As the catalyst type and Platinum Group Metals (PGM) loading are unknown for the EATS used in the experimental tests, this data is taken from the literature [10]. Cu-Zeolite is used as the catalyst material in this model. The rate of reactions and activation energies, along with the chemical reactions, are also used from the same reference. The physical properties of the exhaust gas including composition, and temperature were matched to the engine experimental data. The

TABLE 1: SPECIFICATIONS OF HEAVY-DUTY COMMERCIAL TRUCK

Vehicle model	2016 Freightliner CA125DC
Engine model	Detroit Diesel DD15
Emission certification	GHG14, EPA2010
$H_2$ -DF system	Hydra Energy Co-combustion
Aftertreatment system	DOC + DPF + SCR
No. of Cylinders	6
Displacement (L)	14.8

simulated mass flowrate was scaled relative to the tested engine system to match SCR space velocity.

A sensitivity analysis of this model was performed for parameters including SCR temperature, DEF dosing, NO/NO<sub>2</sub> ratio and space velocity. The sensitivity analysis highlighted important parameters that are required for the modeling of SCR for H<sub>2</sub>-DF engines.

### III. RESULTS AND DISCUSSION

#### A. Experimental Data Analysis

Experimental measured data from H<sub>2</sub>-DF engine was analyzed for emissions as well as the temperature of EATS. The results for emissions and temperatures are discussed separately below:

##### Temperatures:

The thermal profiles of EATS vary across components due to their distinct operational roles, such as oxidation and reduction reactions, and are influenced by the operating condition and whether H<sub>2</sub>-DF or conventional diesel operation. Each component needs a specific stabilization time to give its optimal performance, with the overall aftertreatment system achieving stability subsequent to the stabilization of engine parameters. During the stabilization phase, the performance of these components is unpredictable due to ongoing chemical reactions influenced by multiple factors.

The stabilization duration differs among components; for instance, the DOC typically stabilizes within 5 minutes. Being further downstream, the inlet conditions to the SCR do not stabilize until the DOC and DPF have reached constant temperatures. Furthermore, the active control over DEF dosing, the storage of ammonia on the catalytic sites, and the complex reduction-reaction chemistry lead to stabilization periods of 8 to 10 minutes for the tested conditions.

The aftertreatment system takes several minutes to reach stable temperatures after the engine operation stabilizes, as illustrated in Fig. 2 & 3. The time series in the figures start when engine operation was stable, defined by constant intake air flow rate, intake and engine exhaust system temperatures and pressures, and constant output torque. In the diesel scenario (Fig. 2), the DOC temperature stabilizes after 75 seconds, while the SCR inlet and outlet temperatures initially decline before reaching the stabilization point (at 340 seconds). Conversely, in the hydrogen-diesel dual-fuel (H<sub>2</sub>-DF) case (Fig. 3), both DOC and SCR inlet and outlet temperatures rise initially and then stabilize. DOC became stable at 240 seconds, while SCR stabilized at 420 seconds. These differences are primarily a function of the historical condition of the EATS based on the preceding test points.

##### Stabilization Phase:

During unstable EATS conditions, the SCR inlet temperature slightly exceeds the DOC outlet temperature for diesel-only operation. In contrast, for H<sub>2</sub>-DF, a temperature drop is observed between the DOC outlet and SCR inlet, suggesting an impact of hydrogen on Diesel Particulate Filter (DPF) operation. Additionally, in the H<sub>2</sub>-DF scenario, the SCR outlet temperature is lower than its inlet temperature, whereas, for diesel, the SCR

outlet temperature surpasses the inlet temperature. This difference is more prominent in unstable conditions but becomes negligible as the system stabilizes. This suggests that the reaction kinetics inside the SCR are changed, affecting the thermal conditions. Understanding these thermal behaviors is crucial for optimizing the performance and efficiency of aftertreatment systems in varying fuel contexts.

##### Stable Condition:

The temperature outlet from the DOC is significantly higher in the case of H<sub>2</sub>-DF than for conventional diesel. As the DOC inlet temperature for both fuels is almost the same and follows the exhaust gas temperature trend, there is a significant rise in temperature in H<sub>2</sub>-DF in the DOC. This is likely due to the exothermic oxidation reaction of unreacted H<sub>2</sub>. This high temperature in the DOC outlet results in higher temperatures in the DPF (Fig. 1) and at the inlet to the SCR (Fig 2 & 3). The temperature change across the SCR is small in stable conditions for both diesel and H<sub>2</sub>-DF cases.

##### Emissions:

The emissions formed from diesel fuel combustion include CO<sub>2</sub>, CO, NO<sub>x</sub> (NO and NO<sub>2</sub>), H<sub>2</sub>O molecules and unreacted Hydrocarbons (HC). In the H<sub>2</sub>-DF case, there would be some hydrogen in the exhaust gas. The time series data of onboard NO<sub>x</sub> sensors measuring concentrations at both the inlet and outlet of the EATS for diesel and H<sub>2</sub>-DF operations is shown in Fig. 2 & 3, respectively. These values were compared with the data measured using FTIR and found to be very close to FTIR values, as shown in Fig. 3.

The average engine out emissions entering the EATS are different for diesel and H<sub>2</sub>-DF, as shown in Fig. 4. Once the engine operation stabilized, these concentrations were constant throughout the testing, including EATS stabilization phase and steady-state phases. Engine-out emissions of CO and CO<sub>2</sub> were higher for diesel than for H<sub>2</sub>-DF. For the same reason, H<sub>2</sub>O content increases in H<sub>2</sub>-DF. Engine-out NO<sub>x</sub> emissions, both for the H<sub>2</sub>-DF and diesel cases, remained relatively stable throughout the test and are comparable at this operating point at approximately 200 ppm.

##### Stabilization Phase:

For diesel only operation, SCR-out NO<sub>x</sub> emissions increase initially (Fig. 2 (bottom)). This correlates with a decrease in SCR outlet temperature (Fig. 2 (top)). This demonstrates the SCR performance's dependence on system temperatures that respond relatively slowly to changes in exhaust stream conditions. Initially, the SCR-out temperature was higher due to the previous condition and test point, then due to a change in the test point and inlet conditions, both SCR inlet and outlet temperature started decreasing. This results in a reduction in SCR performance as well as further reduced SCR out temperature, thus increasing the NO<sub>x-out</sub> emissions at 150 seconds while the NO<sub>x-in</sub> was almost constant. After ~300 s, the SCR performance stabilized at approximately 85-90% conversion. This relatively low value is a direct result of the low temperature in the SCR (~250°C).

For the H<sub>2</sub>-DF case, NO<sub>x-out</sub> initially increased, corresponding to a reduction in SCR inlet and outlet temperature

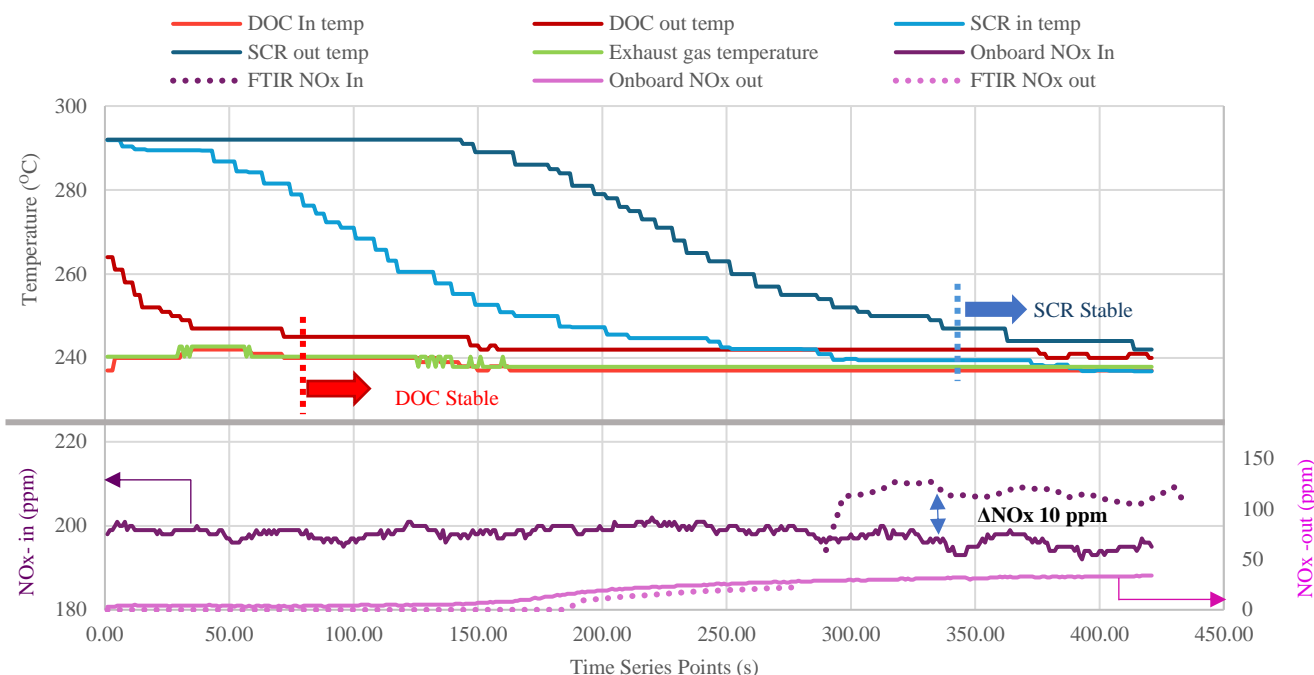


Figure 2: Time series data of DOC and SCR temperature, onboard NO<sub>x</sub> sensors and FTIR NO<sub>x</sub> for Diesel-only operation during EATS stabilization phase at 1200 RPM, 200 Nm shaft torque and an exhaust mass flowrate of 430 kg/hr.

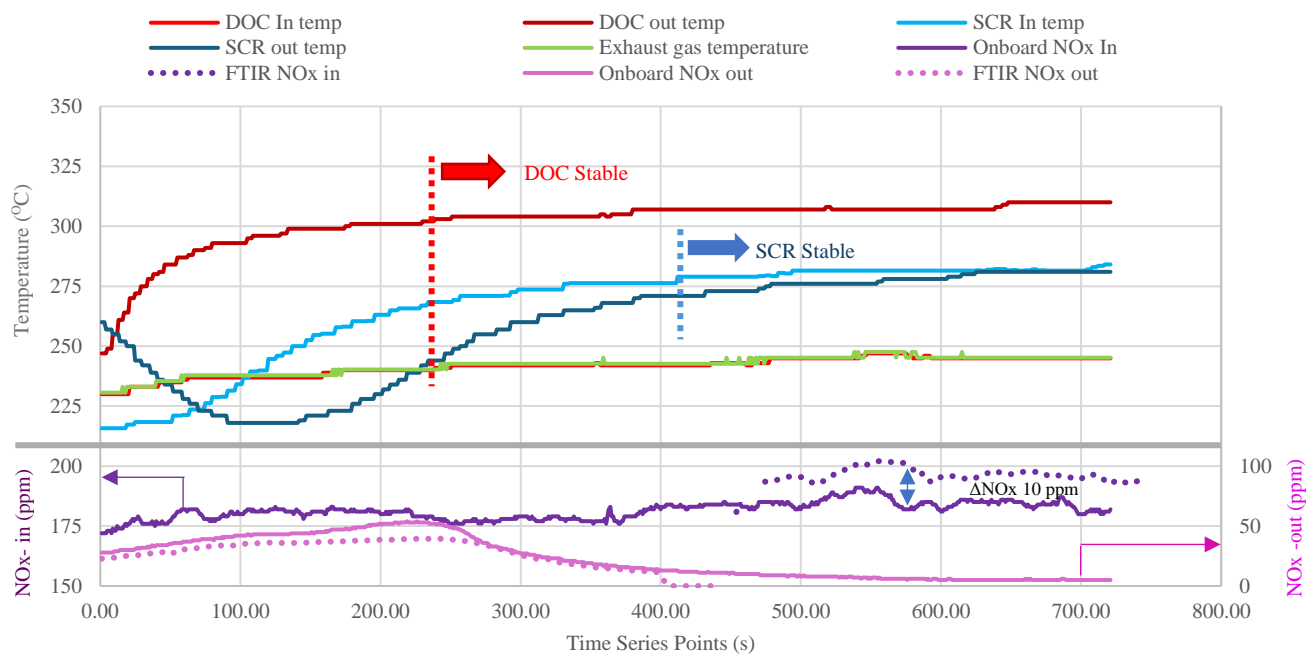


Figure 3: Time series data of DOC and SCR temperature, onboard NO<sub>x</sub> sensors and FTIR NO<sub>x</sub> for H<sub>2</sub>-DF operation during EATS stabilization phase at 1200 RPM, 200 Nm shaft torque and an exhaust mass flowrate of 470 kg/hr.

(Fig 3). The increase in DOC outlet temperature, attributed to the oxidation of H<sub>2</sub> in the H<sub>2</sub>-DF case, increases the temperatures in the subsequent components. The time delay between the high DOC outlet temperature (~100s) and the SCR inlet temperature (~250 s) is due to the time needed to increase the temperature in the DPF (see Fig. 1). Once the higher temperature from the DOC

propagates to the SCR, the performance increases significantly, at ~ 280 seconds.

Notably, when SCR is unstable with low temperatures, overall NO<sub>x</sub> emissions downstream of the SCR are slightly higher for H<sub>2</sub>-DF compared to diesel, despite overall very similar inlet NO<sub>x</sub> concentrations. It is possible that the increased

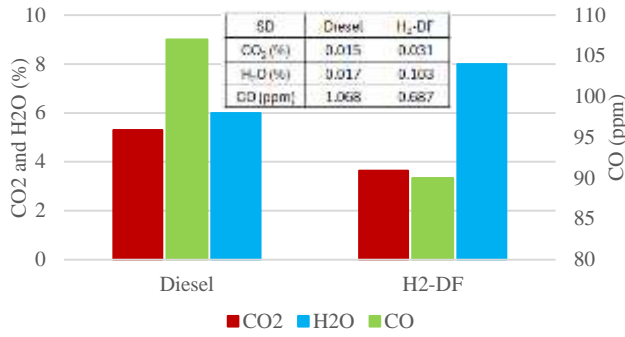


Figure 4: Pre-aftertreatment emissions for Diesel and H<sub>2</sub>-DF at 1200 RPM and 200 Nm shaft torque

moisture content associated with H<sub>2</sub>-DF combustion may adversely affect SCR catalyst performance, leading to reduced NO<sub>x</sub> conversion efficiency at lower loads. [11] [12] This can be offset by higher temperatures from oxidation of H<sub>2</sub> in the DOC, but this depends on the stabilization phase which appears to be longer in H<sub>2</sub>-DF as compared to diesel for this operating point.

#### Stable condition:

The SCR showed high NO<sub>x</sub> conversion during its stable condition for both diesel and H<sub>2</sub>-DF operation, but the overall NO<sub>x</sub> conversion efficiency (DeNO<sub>x</sub>) was higher for H<sub>2</sub>-DF as compared to diesel fuel as shown in Fig. 5. H<sub>2</sub> oxidation in the DOC increases the DOC outlet temperature, increasing the SCR inlet temperature. If any H<sub>2</sub> is left after passing through the DOC and DPF, it will likely participate in the NO<sub>x</sub> reduction reactions – which could further increase NO<sub>x</sub> reduction rates. [8] [9]

Other than these emissions, the NO and NO<sub>2</sub> ratio also differs for both fuels, as indicated in Fig. 5. The average NO<sub>x</sub>-in (NO and NO<sub>2</sub>) for both stable and unstable phases are used for the comparison. Engine-out, NO is about 80% of total NO<sub>x</sub> for diesel, while for H<sub>2</sub>-DF, it is only ~40% of the total. Other studies also showed the same observation about NO and NO<sub>2</sub> ratio at low loads, but this difference becomes less prominent at higher load. [6] Post-EATS NO<sub>x</sub> is almost entirely NO<sub>2</sub> based on the FTIR results collected during the stabilization phase; This may be due to the limitation of FTIR for detecting NO below a certain threshold. The actual concentration values shown in Fig. 5 for NO<sub>x</sub>-out are from the on-board sensor values during the stable operating phase only.

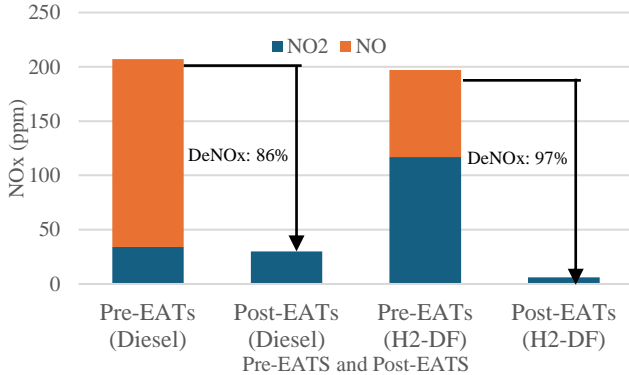


Figure 5: Engine-out (pre-EATS) NO<sub>2</sub>/NO breakdown from FTIR recorded while engine operation was stable, and NO<sub>x</sub> out from on-board sensors during SCR stable phase (shown on Figure 2 & 3).

#### B. Sensitivity Analysis

Predictive modeling of EATS components, particularly the SCR system, provides additional insight for this assessment. This model requires detailed inputs, including chemical kinetics, catalyst properties, exhaust gas characteristics, and composition.

A sensitivity analysis was conducted on the diesel-based SCR model, focusing on parameters such as temperature, ammonia dosing, NO/NO<sub>x</sub> ratio, and mass flow rate, as shown in Fig.6. The key findings from this analysis are as follows:

- **Temperature:** NO<sub>x</sub> conversion efficiency (DeNO<sub>x</sub>) presents a direct relationship with SCR temperature. It reaches a maximum value within an optimal temperature range and then starts declining at higher temperatures after an optimal point. The modelling results are well aligned with the data available in the literature on the temperature effect.
- **Ammonia Dosing:** An increase in NH<sub>3</sub> dosing also relates with improved NO<sub>x</sub> conversion. However, after a certain limit, excessive dosing does not further improve NO<sub>x</sub> reduction and may lead to increased ammonia slip.
- **NO/NO<sub>2</sub> Ratio:** An equimolar NO/NO<sub>2</sub> ratio results in higher DeNO<sub>x</sub> efficiency at consistent ammonia dosing rates. This effect is more prominent at lower temperatures.

The analysis also highlights the significance of ammonia storage on the catalyst surface. During engine operation, Diesel Exhaust Fluid (DEF) dosing occurs to achieve the desired NO<sub>x</sub> conversion. Not all dosed ammonia is immediately utilized; a portion is adsorbed onto catalyst sites, contributing to a reservoir of stored ammonia. This stored ammonia facilitates prompt NO<sub>x</sub> reduction in subsequent engine cycles, provided other conditions are optimal.

The initial NO<sub>x</sub> conversion rate is influenced by the extent of ammonia coverage on the catalyst, as illustrated in Fig 7. Conversion rates increase with ammonia coverage up to

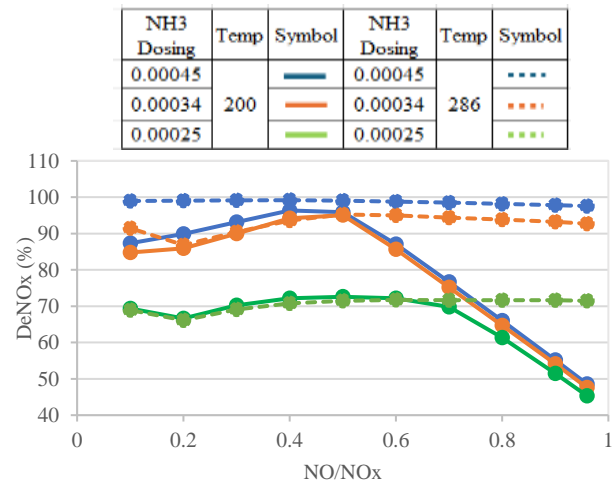


Figure 6: NO<sub>x</sub> conversion for different NO/NO<sub>x</sub> ratios at different temperatures and dosing rates (Solid lines represent 200°C temperature and dashed lines represent 286°C temperature; different color represents different ammonia dosing)



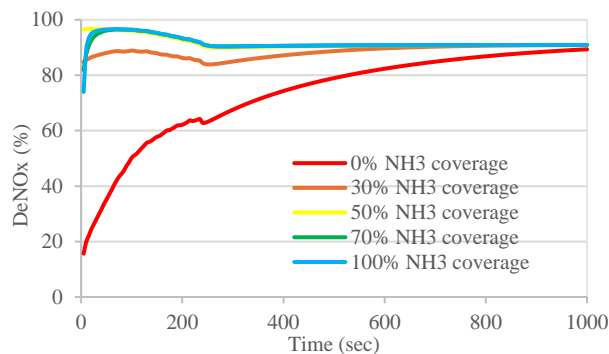


Figure 7: Modelling results of NO<sub>x</sub> conversion at different coverages for Diesel SCR at 25% engine load

approximately 50%, beyond which the rate may slightly decline. If hydrogen is added to the system, it will also get adsorbed on the catalyst surface inside the SCR thus reducing the NH<sub>3</sub> adsorption and impacting the SCR's performance.

The amount of unburned H<sub>2</sub> emitted from the engine that reaches the SCR is uncertain. In most conditions, much of the H<sub>2</sub> will oxidize in the DOC and DPF, leaving less H<sub>2</sub> for reaction inside the SCR. However, H<sub>2</sub> in the DOC will tend to promote the conversion of NO into NO<sub>2</sub>. Higher NO<sub>2</sub>/NO ratios (lower NO/NO<sub>x</sub> in Fig. 6) can help to promote SCR at low temperatures, even if no H<sub>2</sub> is directly reacting in SCR. [7] The behavior of EATS components can be interpreted through the model using experimentally measured data and comparing the results. Modeling and sensitivity analysis, along with experimental data, are important to understand the performance of H<sub>2</sub>-DF aftertreatment system.

#### IV. CONCLUSION

Integrating hydrogen into existing diesel-powered heavy-duty vehicles influences both emission and the performance of exhaust aftertreatment systems (EATS). The presence of H<sub>2</sub> impacts the system component temperatures in the EATS. In H<sub>2</sub>-DF operation, the DOC shows temperature increase. The EATS takes 5 to 8 minutes to stabilize, with the SCR being the slowest to reach stable conditions. Sensitivity analyses of diesel-based SCR models highlighted the impact of SCR operating temperature, the ratio of NO to total NO<sub>x</sub>, the amount of ammonia stored on the catalyst, and the quantity of Diesel Exhaust Fluid (DEF) dosed on its performance.

A summary of the key points concluded from this study is as below:

1. Hydrogen addition effectively lowers CO and CO<sub>2</sub> emissions at low load but increases the water content, which may impact the SCR performance.
2. Temperature evaluation shows a higher temperature increase across the DOC for H<sub>2</sub>-DF, unlike in diesel operation
3. SCR conversion efficiency is sensitive to the initial conditions (i.e. the urea dosing and the NH<sub>3</sub> stored on the catalyst, as demonstrated by the modelling work) as well as the stabilization time.

4. Under these conditions, tailpipe NO<sub>x</sub> emissions were low for both diesel and H<sub>2</sub>-DF operation despite the low-load.
5. At low load conditions, the stabilization time for the EATS is long: especially for the SCR. While in-service engines will rarely be fully stable, the system must be allowed to stabilize to reliably assess the implications of H<sub>2</sub>-DF operation on SCR performance and tailpipe NO<sub>x</sub> emissions.

These findings emphasize the need to achieve a balance between emission reductions and aftertreatment system performance.

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#### REFERENCES

- [1] Archit Srinivasacharya Ayodhya & Kumar Gottkere Narayanappa, "An overview of after-treatment systems for diesel engines," Environmental Science and Pollution Research, 2018.
- [2] Jaltest Diagnostics, Everything about Aftertreatment Systems in the Commercial Vehicle industry. [Online] Available: [https://www.jaltest.com/Content/Catalogos/Catalogos/docu mento tecnico\\_EN\\_883.pdf](https://www.jaltest.com/Content/Catalogos/Catalogos/docu mento tecnico_EN_883.pdf) [Assessed on 19 December 2024]
- [3] Maroa Semakula and Prof Freddie Inambao, "The Formation, Effects and Control of Oxides of Nitrogen in Diesel Engines," International Journal of Applied Engineering Research, 2018.
- [4] Brent Delfel, Introduction to diesel exhaust aftertreatment, 2022. [Online] Available: <https://www.vehicleservicepros.com/service-repair/undercar/article/21259772/introduction-to-diesel-exhaust-aftertreatment> [Assessed on: 18 April 2023]
- [5] Louise Olsson, Hanna Sjövall, Richard J. Blint, "A kinetic model for ammonia selective catalytic reduction over Cu-ZSM-5," Applied Catalysis, 2008
- [6] M. Guan, J. Rochussen, P. Steiche, N. Sapkota, R. Farzam, G. McTaggart-Cowan, S.N. Rogak, P. Kirchen, "Characterizing hydrogen-diesel dual-fuel performance and emissions in a commercial heavy-duty diesel truck," International Journal of Hydrogen Energy Volume 86, 11 October 2024, Pages 1085-1096
- [7] J.M. Herreros, S.S. Gill, I. Lefort, A. Tsolakis, P. Millington, E. Moss, "Enhancing the low temperature oxidation performance over a Pt and a Pt-Pd diesel oxidation catalyst," Applied Catalysis B: Environmental, Volume 147, 5 April 2014, Pages 835-841
- [8] A. Abu-Jrai (Birmingham University), "The effect of H<sub>2</sub> and CO on the selective catalytic reduction of NO<sub>x</sub> under real diesel engine exhaust conditions over Pt/Al<sub>2</sub>O<sub>3</sub>," International Journal of Hydrogen Energy, 2007.
- [9] Alghamdi, "Hydrogen Selective Catalytic Reduction of Nitrogen Oxide on Pt-and Pd-Based Catalysts for Lean-Burn Automobile Applications," SAE International, 2020.
- [10] Atul Pant and Steven J. Schmieg, Kinetic Model of NO<sub>x</sub> SCR Using Urea on Commercial Cu-Zeolite catalyst, Industrial and Engineering Chemistry Research, 2011
- [11] Rongrong Gui, Qinghua Yan, Tianshan Xue, Yanshan Gao, Yuran Li, Tingyu Zhu, Qiang Wang, "The promoting/inhibiting effect of water vapor on the selective catalytic reduction of NO<sub>x</sub>," Journal of Hazardous Materials, Volume 439, 5 October 2022.
- [12] G. Landi, L. Lisi, R. Pirone, G. Russoc, M. Tortorelli, "Effect of water on NO adsorption over Cu-ZSM-5 based catalysts", Catalysis Today, 2012